

Name _____

WRITE THE
MOLECULE

$$\begin{array}{ccccccc} & \text{H} & & \text{H} & & \text{H} & \\ & | & & | & & | & \\ \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\ & | & & | & & | & \\ & \text{H} & & \text{H} & & \text{H} & \end{array}$$

C_3H_8
Propane

$$\begin{array}{ccccccc}
 & \text{H} & & \text{H} & & \text{H} & \\
 & | & & | & & | & \\
 \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\
 & | & & | & & | & \\
 & \text{H} & & \text{H} & & \text{H} & \\
 & & & & & | & \\
 & & & & & \text{H} & - \text{C} - \text{H} \\
 & & & & & | & \\
 & & & & & \text{H} &
 \end{array}$$

C_4H_{10}
BUTANE

$$\begin{array}{ccccccc} & \text{H} & & \text{H} & & \text{H} & & \text{H} \\ & | & & | & & | & & | \\ \text{H} & - \text{C} & = & \text{C} & - & \text{C} & - & \text{C} & - & \text{H} \\ & & & & & | & & | \\ & & & & & \text{H} & & \text{H} \end{array}$$

C₄H₈
BUTENE

$$\begin{array}{ccccccc} & \text{H} & & \text{CH}_3 & & \text{H} & \\ & | & & | & & | & \\ \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\ & | & & | & & | & \\ & \text{H} & & \text{H} & & \text{H} & \end{array}$$
$$\begin{array}{c} \text{C}_4\text{H}_{10} \\ \text{H} \quad \text{H} \\ \text{2-methylpropane} \end{array}$$
$$\begin{array}{ccccccc} & & & \text{H} & & & \\ & & & | & & & \\ \text{H} & - & \text{C} \equiv & \text{C} & - & \text{C} & - & \text{H} \\ & & & | & & & \\ & & & \text{H} & & & \end{array}$$

C_3H_4
Propyne

7.

NOTE: CORRECTION

$$\begin{array}{ccccccc} & \text{H} & & \text{H} & & \text{H} & & \text{H} & & \text{H} \\ & | & & | & & | & & | & & | \\ \text{H} & - \text{C} & - & \text{C} & = & \text{C} & - & \text{C} & - & \text{C} & - & \text{H} \\ & | & & & & & & | & & | \\ & \text{H} & & & & & & \text{H} & & \text{H} \end{array}$$

C_5H_{10}

Pentene

$$\begin{array}{ccccccc} & \text{H} & & \text{H} & & \text{H} & \\ & | & & | & & | & \\ \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - & \text{C} & - & \text{C} & - & \text{H} \\ & | & & | & & | & & | & & | & \\ & \text{H} & & \text{H} & & \text{H} & & \text{H} & & \text{H} & \end{array}$$

C₆H₁₄
isomer to
(Hexane)
2-methylpentane

8. C_9H_{20}
3,3-diethylpentane

```
      H
      |
H - C - C - C - C - C - H
      |   |   |   |
      H   H   H   H
      |   |   |   |
      H   H   H   H
```

Naming And Drawing Hydrocarbons Worksheet

Steve Lewis

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Naming And Drawing Hydrocarbons Worksheet:

Basics of Industrial Hygiene Debra Nims, 1999-01-28 This book provides environmental technology students with an enjoyable way to quickly master the basics of industrial hygiene Like all the books in the critically acclaimed Preserving the Legacy series it follows a rapid learning modular format featuring learning objectives summaries chapter end reviews practice questions and skill building classroom activities Throughout the text sidebars highlight critical concepts and more than 90 high quality line drawings photographs and diagrams help to clarify concepts covered Author Debra Nims begins with a fascinating historical overview of the art and science of industrial hygiene followed by a concise review of key concepts and terms from biology and toxicology She then offers in depth practical coverage of Identifying hazards or potential hazards Sampling and workplace evaluations Hazard control Toxicology occupational health and occupational health standards Airborne hazards Dermatoses and contact hazards Fire and explosion hazards Occupational noise Radiation Temperature extremes Repetitive use traumas With its comprehensive coverage and quick reference format Basics of Industrial Hygiene is also a handy refresher and working reference for practicing environmental technicians and managers

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