

Matter

Study Guide & Review Worksheets

Matter

Terms to Know

- **Matter**, Anything that has mass and takes up space
- **Mass**, The amount of matter in an object
- **Atoms**, Tiny particles that matter can be subdivided into

Mixtures and Solutions

Mixtures

A mixture is when two or more substances are combined. They do not lose their characteristics.

Examples



soil



concrete



mud puddle

Solutions

Solutions are a special type of mixture in which one substance is uniformly dissolved in a liquid.

Examples



sugar water



soda



salt water

Temperature and Phase Changes

Matter can change from one state to another when the temperature increases or decreases.

When the temperature increases, a solid will change to a liquid and then to a gas.

Example:

When a pot of water is put on a hot stove, it boils. The liquid water changes to water vapor.



When the temperature decreases, a gas will change to a liquid and then to a solid.

Example:

When liquid freeze pops are put into a freezer, they become solid.



Mass and Phase Changes

Matter does not gain or lose mass during phase changes.

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Terms to Know

Anything that has mass and takes up space. Matter in an object can be subdivided into atoms.

Mixtures and Solutions

Solutions

Solutions are a special type of mixture in which one substance is uniformly dissolved in a liquid.

List 3 examples:

Temperature and Phase Changes

Matter can change from one state to another when the temperature increases or decreases.

When the temperature increases, a solid will change to a liquid and then to a gas.

Example:

Mass and Phase Changes

Matter does not gain or lose mass during phase changes.

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Matter Study Guide

James E. Brady, John R. Holum



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Glencoe Chemistry Glencoe/McGraw-Hill,2002 *A Study Guide for Chemistry, Matter and the Universe* Ellen A. Keiter,Richard L. Keiter,1976 Chemistry, Study Guide Neil D. Jespersen,James E. Brady,Alison Hyslop,2011-05-03 Chemistry The Study of Matter and Its Changes Sixth Edition will provide the necessary practice support and individualised instruction that ensures success in the General Chemistry course This text provides the forum for problem solving and concept mastery of chemical phenomena that leads to proficiency and success in the General Chemistry course This edition will continue a molecular basis of chemistry tradition but in a manner that overtly and repeatedly reinforces the way properties at the molecular level are related to properties we observe at the macroscopic level The unique chemical tools approach employed in this book provides a way of thinking that helps those students develop the ability to analyse and solve both mathematical and conceptual problems This text follows the successful three step approach described as ANALYSIS SOLUTION and IS THE ANSWER REASONABLE This encourages the student to think about the problem before attempting to solve it then working through the solution and finally asking the important question Does the answer make sense There are problem sets called Bringing It Together that contain problems which require students to bring together concepts from two or more of the preceding chapters This reinforces learned concepts and builds concept mastery *Chemistry, Study Guide* James E. Brady,John R. Holum,1996-03-01 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving **Resources in Education** ,1984 Study Guide to accompany Chemistry: Matter and Its Changes, 4th Edition James E. Brady,Frederick A. Senese,2004-01-12 Work more effectively and gauge your progress as you go along This Study Guide is designed to accompany Brady Senese s Chemistry Matter Its Changes 4th Edition It contains learning objectives chapter reviews examples and self test questions Brady Senese s Chemistry Matter Its Changes 4th Edition is a reader friendly textbook that makes the content accessible without sacrificing either breadth or depth of coverage The text s informal writing style emphasis on problem solving and state of the art media package make this book an ideal fit for readers of various backgrounds and abilities The 4th edition welcomes new co author Fred Senese the architect of the most visited general chemistry website Together Jim Brady and Fred Senese offer accurate lucid and interesting explanations of the basic concepts of chemistry as well as comprehensive coverage and aid to readers in developing problem solving skills

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



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