



सामान्य निर्देश :

निम्नलिखित निर्देशों को बहुत सावधानी से पढ़िए और उनका साख्ती से पालन कीजिए :

- यह प्रश्न-पत्र चार खण्डों में विभाजित किया गया है – क, ख, ग और घ ।
- इस प्रश्न-पत्र में 37 प्रश्न हैं । सभी प्रश्न अनिवार्य हैं ।
- खण्ड क – प्रश्न संख्या 1 से 20 तक अति लघु-उत्तरीय प्रश्न हैं, प्रत्येक प्रश्न 1 अंक का है ।
- खण्ड ख – प्रश्न संख्या 21 से 27 तक लघु-उत्तरीय प्रश्न हैं, प्रत्येक प्रश्न 2 अंकों का है ।
- खण्ड ग – प्रश्न संख्या 28 से 34 तक दीर्घ-उत्तरीय प्रकार के प्रश्न हैं, प्रत्येक प्रश्न 3 अंकों का है ।
- खण्ड घ – प्रश्न संख्या 35 से 37 तक भी दीर्घ-उत्तरीय प्रकार के प्रश्न हैं, प्रत्येक प्रश्न 5 अंकों का है ।
- प्रश्न-पत्र में कोई समग्र विकल्प नहीं है । तथापि, एक-एक अंक के दो प्रश्नों में, दो-दो अंकों वाले दो प्रश्नों में, तीन-तीन अंकों वाले एक प्रश्न में तथा पाँच-पाँच अंकों वाले तीनों प्रश्नों में आंतरिक विकल्प दिया गया है । ऐसे प्रश्नों में केवल एक ही विकल्प का उत्तर दीजिए ।
- इसके अतिरिक्त, आवश्यकतानुसार, प्रत्येक खण्ड और प्रश्न के साथ यथोचित निर्देश दिए गए हैं ।
- कैल्कुलेटर्स अथवा लॉग टेबलों के प्रयोग की अनुमति नहीं है ।
- जहाँ आवश्यक हो, आप निम्नलिखित भौतिक नियतांकों के मानों का उपयोग कर सकते हैं :

$$c = 3 \times 10^8 \text{ m/s}$$

$$h = 6.63 \times 10^{-34} \text{ Js}$$

$$e = 1.6 \times 10^{-19} \text{ C}$$

$$\mu_0 = 4\pi \times 10^{-7} \text{ T m A}^{-1}$$

$$\epsilon_0 = 8.854 \times 10^{-12} \text{ C}^2 \text{ N}^{-1} \text{ m}^{-2}$$

$$\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ N m}^2 \text{ C}^{-2}$$

$$\text{इलेक्ट्रॉन का द्रव्यमान (m_e)} = 9.1 \times 10^{-31} \text{ kg}$$

$$\text{न्यूट्रॉन का द्रव्यमान} = 1.675 \times 10^{-27} \text{ kg}$$

$$\text{प्रोटॉन का द्रव्यमान} = 1.673 \times 10^{-27} \text{ kg}$$

$$\text{आवोगाद्रो संख्या} = 6.023 \times 10^{23} \text{ प्रति ग्राम मोल}$$

$$\text{बोल्ट्ज़मान नियतांक} = 1.38 \times 10^{-23} \text{ JK}^{-1}$$

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Guillaume Favre



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