

## Chemistry – Unit 5 – Worksheet 2

Show work. Label set-ups and answers with units. *Box answers.*

1. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?
2. Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?
3. What mass of water would you need to have 15.0 moles of  $\text{H}_2\text{O}$ ?
4. One box of Morton's Salt contains 737 grams. How many moles of sodium chloride,  $\text{NaCl}$ , is this?
5. A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). Sodium bicarbonate has the chemical formula,  $\text{NaHCO}_3$ . How many grams should the chef mass out?
6. Lead (II) chromate,  $\text{PbCrO}_4$ , was used as a pigment in paints. How many moles of  $\text{PbCrO}_4$  are in 75.0 g of lead (II) chromate? How many atoms of oxygen are present?

# Living By Chemistry Worksheets Unit 2

**Holt Rinehart & Winston, Holt,  
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