

AP Chemistry Practice Test Chapter 8 and 9

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ____ 1. Which pair of atoms should form the most polar bond?
a. F and B
b. C and O
c. F and O
d. N and F
e. B and N
- ____ 2. Which pair of ions should form the ionic lattice with the highest energy?
a. Na^{1+} and Br^{1-}
b. Li^{1+} and F^{1-}
c. Cs^{1+} and F^{1-}
d. Li^{1+} and O^{2-}
e. K^{1+} and F^{1-}
- ____ 3. The electrostatic force of attraction is greatest in which compound?
a. BaO
b. MgO
c. CaS
d. MgS
e. CaO
- ____ 4. Which molecule has the weakest bond?
a. CO
b. O_2
c. NO
d. N_2
e. Cl_2
- ____ 5. For which species are octet resonance structures necessary and sufficient to describe the bonding satisfactorily?
a. BCl_3
b. SO_2
c. CO_2
d. BeF_2
e. SO_4^{2-}
- ____ 6. How are the bonding pairs arranged in the best Lewis structure for ozone?
a. $\text{O}-\text{O}-\text{O}$
b. $\text{O}=\text{O}-\text{O}$
c. $\text{O}=\text{O}=\text{O}$
d. $\text{O}=\text{O}=\text{O}$
e. $\text{O}=\text{O}-\text{O}$
- ____ 7. Which species has the shortest bond length?
a. CN^{1-}
b. O_2
c. SO_2
d. SO_3
e. CO_2
- ____ 8. Which species requires the least amount of energy to remove an electron from the outermost energy level?
a. Na^{1+}
b. He
c. F^{1-}
d. O^{2-}
e. Mg^{2+}
- ____ 9. Which species has a valid non-octet Lewis structure?
a. GeCl_4
b. SiF_4
c. CO_3^{2-}
d. SeCl_4
e. CO_3^{2-}

Matter Chemistry Pre Ap Practice Tests

John T. Moore, Richard H. Langley



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Barron's AP Chemistry Neil D. Jespersen, 2012-02-01 Extensive test preparation for the AP Chemistry exam includes Six practice AP exams three diagnostic tests and three full length practice exams All questions answered and explained A comprehensive subject review covering the structure of matter chemical bonding states of matter physical chemistry chemical reactions and all other test topics Study tips and test taking strategies **AP Chemistry** P. E. Dumas, R. M. Fikar, J. W. Samples, W. C. Uhland, 2008-01-22 REA Real review Real practice Real results Get the college credits you deserve AP CHEMISTRY 10th Edition Completely aligned with today's AP exam Are you prepared to excel on the AP exam Set up a study schedule by following our results driven timeline Take the first practice test to discover what you know and what you should know Use REA's advice to ready yourself for proper study and success Practice for real Create the closest experience to test day conditions with 6 full length practice tests Chart your progress with full and detailed explanations of all answers Boost your confidence with test taking strategies and experienced advice Sharpen your knowledge and skills The book's full review features coverage of all AP Chemistry main topic areas such as solutions stoichiometry kinetics and thermodynamics as well as all subject areas found on the official exam including the structure of matter the states of matter chemical reactions and descriptive chemistry Smart and friendly lessons reinforce necessary skills Key tutorials enhance specific abilities needed on the test Targeted drills increase comprehension and help organize study Ideal for Classroom Family or Solo Test Preparation REA has provided advanced preparation for generations of advanced students who have excelled on important tests and in life REA's AP study guides are teacher recommended and written by experts who have mastered the course and the test **Barron's AP Chemistry with CD-ROM** Neil D. Jespersen, 2014-02-01 Updated to reflect the brand

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