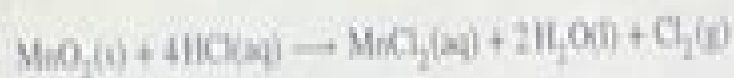


Chlorine can be prepared in the laboratory by the reaction of manganese dioxide with hydrochloric acid,  $\text{HCl(aq)}$ , as described by the chemical equation



How much  $\text{MnO}_2(\text{s})$  should be added to excess  $\text{HCl(aq)}$  to obtain 115 mL  $\text{Cl}_2(\text{g})$  at 25 °C and 775 Torr?

mass of  $\text{MnO}_2$ :

13

g

# Mcquarrie General Chemistry 4th Edition Solutions Manual

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