Semester: 7th	Progr	amme: BS Physics	Roll No.
Course Code: PHY-609	Part Obje	c fines	Crasticut
Time allowed: 30 Minu	Course Title: Statistical Mechanics		Credit Hrs.: 3(3-0) Marks: 29
Q.No.01 Choose the corre	SCE DESCRIPTION		
a) 4 - T wa sieve - o a	nysical quantity A (q. p)	for actual system of interest is	
2 The	b) A - 1 A(q. p) p(q.	p.1)dF c)1 = \int_{-\infty}^{\infty} A(q,p)p(q,p,t)d1	d) A = 1 A(q. p)L(q. p. 1) AT
 The value of γ(C_p/C_*) f a) 1.66 	OF A DIATOMIC GAS	molecule is	
The trajectory of S.H.O	b) 1.4	c) 1.33	D 1
a) Ellipse			
4. In phase space, the total	b) Circle	c) Parabola	d) Spiral
a) Sum of individual	b) Product of individ		
probability	probability		d) Division of individual
5. According to Maxwell-I		individual probability ribution; the average speed \vec{u} of the	probability
a) 8k _B T/zm	b) 8k _B T/am) ^{1/}		
6. If a change is reversible		c) (zm/8k _B T) ^{1/2}	d) πm/8kaT
a) AS _{composite} > 0	b) $\Delta S_{torogram} = 0$	c) $\Delta S_{composite} < 0$ d	
		n can be derived from STATISTIC) both a) & b)
		c) Max Well-Boltzmann	d) (a) and (b)
		e sign when these are interchange	
i) Boson	b)Fermions	c) Photons	d) Negatron
. In case of micro canonic		e of sphere and its thickness shou	
Greater than 1	b) Less than 0	c) Different	d) Same
		energy of an isolated system	o) Same
0) dO ≠ 0	b) dW # 0	c) E = constant	d) All of the mentioned
			d) All of the mentioned
11. A red glass piece is heat	ed until it glows in dark	. The color of the glow will be	d) Violet
	b) Orange	c) Green	
		racterize the thermodynamic state	d) Work
	b) Temperature	c) Pressure	
Which is meaningful ar	rangements for case of a	particles and two boxes, if m in	001-1-01-01-01-0
in box-2			d) nt .(nt-n ₁ !)/ n ₂ !
a) n! .u ₂ !/ n ₁ !	b) all ast ast	c) at .(nt-mit)/ mit	
		erticles n= 3 and compartments C c) 8	d) 1
a) 9	ь) 5	ic gases is (where, I is mean free p	
		c) \(\lambda - 2/3k_BrlpN\)	d) \u20ab=1/2kavlpN
a) \(\lambda = 3/2k_avlpN \)		be used to characterize an	
		c) Open system	d) Isolated system
a) Closed system	b) Particle system	dent events is equal to their	
		c) product	d) Ratio
a) Sum	b) Difference		
18. The system will do w	b) $F = -T \log Z$	c) $F = T \log Z$	d) $F = -T \log A$
a) $F = P \log Z$		at constant pressure and at constan	
	en motar specific nears		
about	13.6	c) 1.41	d) Gibbs Constant
a) Avogardo's number	b) Gas constant		
		() in Energy state ε _κ is given as (T) c) 1 /1+ [exp (ε + μ)/T]	d) 1 /1+ [exp (e - μ)]
a) $1/1 + [\exp(\epsilon - \mu)/T]$	b) 1 /1 - [exp (ε - μ)/	11 c) fexb fe . bb 11	

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- 1. Understanding the eBook Mechanics 6677 2013 Past Paper
 - The Rise of Digital Reading Mechanics 6677 2013 Past Paper
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Mechanics 6677 2013 Past Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mechanics 6677 2013 Past Paper
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Mechanics 6677 2013 Past Paper
 - Personalized Recommendations
 - Mechanics 6677 2013 Past Paper User Reviews and Ratings
 - Mechanics 6677 2013 Past Paper and Bestseller Lists

- 5. Accessing Mechanics 6677 2013 Past Paper Free and Paid eBooks
 - Mechanics 6677 2013 Past Paper Public Domain eBooks
 - Mechanics 6677 2013 Past Paper eBook Subscription Services
 - Mechanics 6677 2013 Past Paper Budget-Friendly Options
- 6. Navigating Mechanics 6677 2013 Past Paper eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Mechanics 6677 2013 Past Paper Compatibility with Devices
 - Mechanics 6677 2013 Past Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mechanics 6677 2013 Past Paper
 - Highlighting and Note-Taking Mechanics 6677 2013 Past Paper
 - o Interactive Elements Mechanics 6677 2013 Past Paper
- 8. Staying Engaged with Mechanics 6677 2013 Past Paper
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mechanics 6677 2013 Past Paper
- 9. Balancing eBooks and Physical Books Mechanics 6677 2013 Past Paper
 - Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection Mechanics 6677 2013 Past Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mechanics 6677 2013 Past Paper
 - Setting Reading Goals Mechanics 6677 2013 Past Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mechanics 6677 2013 Past Paper
 - Fact-Checking eBook Content of Mechanics 6677 2013 Past Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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