

# PHYSICAL SCIENCES P.2



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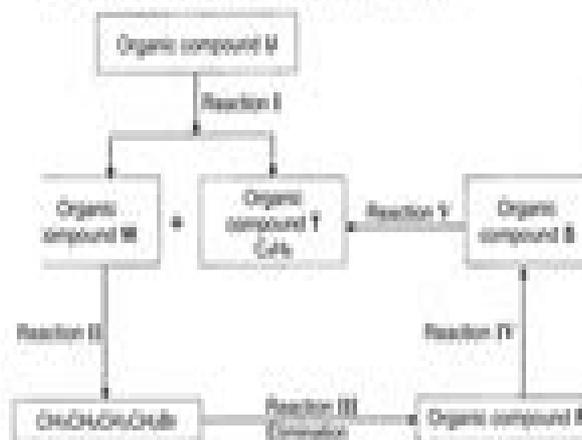
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# MEMO-2024

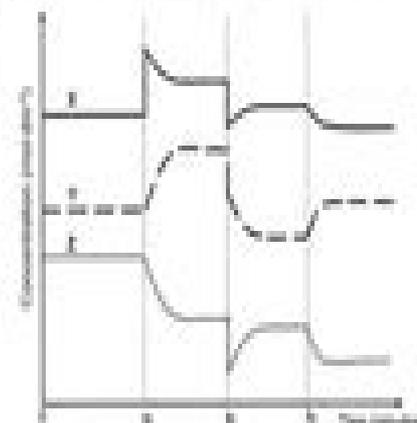
## G.12

### NATIONAL SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



4.5 Write down the TYPE of absorption in reaction II.

4.6 Compounds R and T are positional isomers.

The inorganic reagents shown below are available for reactions IV and V.

Bu	Pb(CH <sub>3</sub> COO) <sub>2</sub>	K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> (H <sub>2</sub> SO <sub>4</sub> )	HBr	K <sub>2</sub>
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4.7 Define for both chemical equilibrium.

Use the graph to answer the questions below.

# Memorandum For Physics Paper 2 November 2014

**Patrick Vollmar**



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