



# basic education

Department:  
Basic Education  
**REPUBLIC OF SOUTH AFRICA**

**SENIOR CERTIFICATE/  
NATIONAL SENIOR CERTIFICATE**

**GRADE 12**

**LIFE SCIENCES P1**

**NOVEMBER 2020(2)**

**MARKS: 150**

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# Matric Life Sciences Ptopics Caps 2014

**J Spring**



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web mar 13 2023 chemical bonds are the attractive forces that hold atoms together in the form of compounds a chemical bond is formed when electrons are shared between two atoms there are three types of bonds covalent bonds polar covalent bonds and ionic bonds the simplest example of bonding is demonstrated by the  $H_2$  molecule

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web jun 14 2022 explain your answer what are the predominant interactions when oppositely charged ions are far apart at internuclear distances close to  $r_0$  very close together at a distance that is less than the sum of the ionic radii several factors contribute to the stability of ionic compounds