

Chemistry - Hall Lab

Purpose

The purpose of this laboratory is to determine the value of copper produced in this experiment as a percentage yield.

Procedure

Step 1

1. Label three pieces of tin 1, 2, and 3.
2. Put 10.0000 g of acid (10.0 g of copper) into the beaker. To do this, pour the solution into the beaker from the top of the beaker. Use a funnel to add copper to the beaker. Use the beaker to add the solution to the beaker.
3. Add about 10.0 g of acid to the beaker. Use a funnel to add the acid to the beaker.
4. Allow it to sit for 10 minutes to 15 minutes.
5. Place the beaker in the copper solution collection. Observe the reaction, record your observations. Place the beaker back in the place designated for your reaction.

Step 2

6. Remove the acid. Remove the beaker from the collection. Remove the acid from the beaker. Remove the beaker from the collection. Place the beaker in a beaker with water. Remove the beaker from the collection. Place the beaker in the drying oven.
7. Remove the acid from the beaker. Remove the beaker from the collection. Remove the acid from the beaker. Remove the beaker from the collection. Place the beaker in a beaker with water. Remove the beaker from the collection. Place the beaker in the drying oven.
8. Remove the acid from the beaker. Remove the beaker from the collection. Remove the acid from the beaker. Remove the beaker from the collection. Place the beaker in a beaker with water. Remove the beaker from the collection. Place the beaker in the drying oven.

Step 3

9. Remove the beaker from the collection. Remove the beaker from the collection. Place the beaker in the drying oven. Remove the beaker from the collection. Place the beaker in the drying oven.

Results

Mass (g) of beaker	1.0000 g
Mass (g) of beaker + copper (10.0000 g)	1.0000 g
Mass (g) of beaker + copper (10.0000 g)	1.0000 g

Modeling Chemistry Nail Lab

AW Rasmussen



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