

## UNIT 8

### QUESTIONS

1. [     /4] Write P if it's physical change, C if it's chemical reaction

Burning a piece of wood .....

Melting chocolate .....

Cooking an egg .....

Heating glass and bending it .....

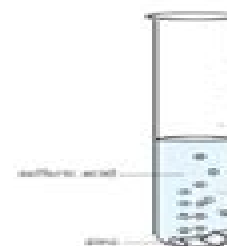
2. [     /3] Zinc is added to sulfuric acid. Answer the questions.

What is the name of the gas ? .....

How do you test for this gas? .....

What are the products in this reaction?

.....



3. [     /3] Magnesium ribbon burns in air.

a. Write the chemical symbol for magnesium .....

b. Name the element that magnesium reacts with when it burns .....

c. Name the compound formed when this element reacts with magnesium

.....

4. [     /4] Write the letter of the reaction that:

a. produces metal oxide      b. a neutralisation reaction      c. a burning reaction

Carbon combines with oxygen to form carbon dioxide	.....
Sodium hydroxide reacts with hydrochloric acid to produce sodium chloride and water	.....
Potassium and water react to form potassium hydroxide and hydrogen	.....
Copper carbonate reacts to produce copper oxide and carbon dioxide	.....

# Modeling Chemistry Test U8 Testa V 2

**R Barnett**



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