Colligative Properties



 As a solute is added to a solvent, what happens to the freezing point and the boiling point of the solution?

A) The freezing point decreases and the boiling point increases.

- B) The freezing point increases and the boiling point increases.
- The freezing point increases and the boiling. point decreases.
- D) The freezing point decreases and the boiling. point decreases.
- Compared to the freezing point of 1.0 M KCl(aq) at standard pressure, the freezing point of 1.0 M. CaCla(aq) at standard pressure is

B) higher

Which aqueous solution of KI freezes at the lowest реширеги писе?

A) 1 mol of KI in 500. g of water

B) 2 mol of KI in 500. g of water (%)

C) I mol of KI in 1000, g of water

D) 2 mol of KI in 1000, g of water

 A I kilogram sample of water will have the highest freezing point when it contains

A) 1 × 10²¹ dissolved particles

(B) 1 × 1017 dissolved particles

C) 1 × 1019 dissolved particles

D) 1 × 10²³ dissolved particles http://s.

 A solution consists of 0.50 mole of CaCl: dissolved in 100, grams of H2O at 25°C. Compared to the boiling point and freezing point of 100. grams of H2O at standard pressure, the solution at standard pressure has

 A) a higher boiling point and a lower freezing. point

B) a lower boiling point and a lower freezing

C) a higher boiling point and a higher freezing.

 D) a lower boiling point and a higher freezing. point

6. Which solution will freeze at the lowest bisimplerature?

A) I mole of sugar in 500 g of water

B) I mole of sugar in 1,000 g of water

C) 2 moles of sugar in 1,000 g of water 2 moles of sugar in 500 g of water

7. Which solution has the highest boiling point at standard pressure?

(A))0.10 M K;PO4(30)

B) 0.10 M KCl(ac)

C) 0.10 M KNO₃(a₀)

D) 0.10 M K2SO4(aq)

 Which concentration of a solution of CH1OH in water has the lowest freezing point?

A) 0.001 M

C) 0.01 M

9. How do the boiling point and freezing point of a solution of water and calcium chloride at standard pressure compare to the boiling point and freezing

point of water at standard pressure?

 A) Both the freezing point and boiling point of the solution are higher.

B) The freezing point of the solution is higher and the boiling point of the solution is lower.

C) The freezing point of the solution is lower and the boiling point of the solution is higher.

 D) Both the freezing point and boiling point of the solution are lower.

10. Compared to a 2.0 M aqueous solution of NaCl at 1 atmosphere, a 3.0 M aqueous solution of NaC1 at I atmosphere has a

A) lower boiling point and a higher freezing.

B) higher boiling point and a higher freezing.

C) lower boiling point and a lower freezing provinct.

higher boiling point and a lower freezing point

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Jiyuan Zhang

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Roadmap to the Regents James Flynn, 2003 If Students Need to Know It It's in This Book This book develops the Earth science skills of high school students It builds skills that will help them succeed in school and on the New York Regents Exams Why The Princeton Review We have more than twenty years of experience helping students master the skills needed to excel on standardized tests Each year we help more than 2 million students score higher and earn better grades We Know the New York Regents Exams Our experts at The Princeton Review have analyzed the New York Regents Exams and this book provides the most up to date thoroughly researched practice possible We break down the test into individual skills to familiarize students with the test's structure while increasing their overall skill level We Get Results We know what it takes to succeed in the classroom and on tests This book includes strategies that are proven to improve student performance We provide content groupings of guestions based on New York standards and objectives detailed lessons complete with skill specific activities three complete practice New York Regents Exams in Physical Setting Earth Science Catalog of **Copyright Entries. Third Series** Library of Congress. Copyright Office,1975 Berkeley at War: The 1960s W.J. Rorabaugh Professor of History University of Washington, 1989-05-04 Berkeley California was the bellwether of the political social and cultural upheaval that made the 1960s a unique period of American history a time when the top down methods of a conservative establishment collided head on with the bottom up grass roots ethos of the civil rights movement and an increasingly well educated and individualistic middle class W J Rorabaugh who attended the graduate school of the University of California at Berkeley in the early 1970s presents a lively and informative account of the events that overtook and changed forever what had once been a guiet conservative white suburb The rise of the Free Speech Movement which gave a voice to disfranchised students the growth and increasing militance of a black community struggling to end segregation the emergence of radicalism and the anti war movement the blossoming of hippie culture with its scorn for materialism and enthusiasm for experimentation with everything from sex and drugs to Eastern philosophies the beginnings of modern day feminism and environmentalism and how all of these coalesced in the explosive conflict over People's Park are traced in a meticulously researched and authoritative narrative At issue was the question of power and the struggle between the establishment and the powerless led to developments that the advocates of a freer society could scarcely have foreseen Ronald Reagan elected governor of California in reaction to the events at Berkeley and Edwin H Meese III who battled against the student movement and People's Park rose to national power in the 1980s without however gaining any popularity in Berkeley where Walter Mondale won 83 percent of the vote in 1984 An invaluable account of its time and place this book anchors the 60s in American history both before and since that colorful decade **Catalogue of Copyright Entries** Library of Congress. Copyright Office, 1918 Catalogue of Title-entries of Books and Other Articles Entered in the Office of the Librarian of Congress, at Washington, Under the Copyright Law ... Wherein the Copyright Has Been Completed by the

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