

<u>Lewis Structure</u>	<u>Molecular Geometry</u>	<u>Bond Polarity</u>	Polar or <u>Non-polar</u>
① SiO_2 $\ddot{\text{O}}=\text{Si}=\ddot{\text{O}}$	Linear	Covalent Bond	Non-polar
② BH_3 $\begin{array}{c} \text{H} - \text{B} - \text{H} \\ \\ \text{H} \end{array}$	Trigonal planar	Covalent Bond	Non-polar
③ NH_3 $\begin{array}{c} \text{H} - \ddot{\text{N}} - \text{H} \\ \\ \text{H} \end{array}$	Trigonal Pyramidal	Covalent Bond	Polar
④ H_2O $\begin{array}{c} \cdot\cdot\text{O}\cdot\cdot \\ \diagup \quad \diagdown \\ \text{H} \quad \text{H} \end{array}$	Tetrahedral	Polar Covalent	Polar
⑤ CO $:\text{C}\equiv\text{O}:$	Linear	Covalent Bond	Polar

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Michael Seilmaier



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have guides and journal lists **Chemical Bonding and Molecular Geometry** Ronald James Gillespie, Paul L. A.
Popelier, 2001 Provides an introduction to models and theories of chemical bonding and geometry as applied to the molecules
of the main group elements This text also elucidates the relationships between these various models and theories It is useful
for courses on chemical bonding in chemistry departments at the senior first year graduate level *Molecular Geometry*
Alison Rodger, Mark Rodger, 2014-05-16 Molecular Geometry discusses topics relevant to the arrangement of atoms The book
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determination of molecular geometry while Chapter 2 discusses the unified view of stereochemistry and stereochemical
changes Chapter 3 covers the geometry of molecules of second row atoms and Chapter 4 deals with the main group elements
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organometallic compounds and transition metal clusters The last chapter tackles the consequences of small local variations
in geometry The text will be of great use to chemists who primarily deal with the properties of molecules and atoms

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the structures of hydrogen halogen and methyl derivatives of the elements in each group some of these molecules are ionic
some polar covalent The survey of molecules whose structures conform to well established trends is followed
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compounds that have not yet been studied The book provides a lucid discussion of a number of models such as the Lewis
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