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ML Morrison

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342g of 20% by mass of Ba(OH)_2 solution specific gravity 1.057 is reacted with 1200 ml of 2M HNO_3 . If the final density of solution is same as pure water, then molarity of the iron in resulting solution which decides the nature of the above solution is

- 0.25M
- 0.5M
- 0.888M
- none of

web what is the formula for converting molarity to molality molality is equal to the product of molarity divided by the density of the solution at that temperature $m = \frac{M}{d}$ which is more important molality or molarity both molality and molarity are concentration units and they are used interchangeably

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