

1. A hypothetical gas that perfectly fits all the assumptions of the kinetic-molecular theory is known as

- |                         |                   |                         |               |
|-------------------------|-------------------|-------------------------|---------------|
| <input type="radio"/> A | an imaginary gas. | <input type="radio"/> B | a real gas.   |
| <input type="radio"/> C | a perfect gas.    | <input type="radio"/> D | an ideal gas. |

2. Which of the following is an assumption of the kinetic-molecular theory of gases?

- |                         |   |                         |  |
|-------------------------|---|-------------------------|--|
| <input type="radio"/> A | Gases consist of closely spaced particles.      | <input type="radio"/> B | Collisions between gas particles are inelastic.                                      |
| <input type="radio"/> C | Gas particles move around in an orderly manner. | <input type="radio"/> D | The temperature of a gas depends on the average kinetic energy of the gas particles. |

3. A liquid can be described as a form of matter that

- |                         |                       |                         |                        |
|-------------------------|-----------------------|-------------------------|------------------------|
| <input type="radio"/> A | has a definite shape. | <input type="radio"/> B | has a definite volume. |
| <input type="radio"/> C | cannot diffuse.       | <input type="radio"/> D | has low density.       |

4. Diffusion is much slower in liquids than in gases because

- |                         |                                       |                         |                                      |
|-------------------------|---------------------------------------|-------------------------|--------------------------------------|
| <input type="radio"/> A | liquids have surface tension.         | <input type="radio"/> B | liquids exist at lower temperatures. |
| <input type="radio"/> C | liquid particles are closer together. | <input type="radio"/> D | liquid particles weigh more.         |

5. The three states of matter of a particular substance, in order of the strength of intermolecular forces from least to greatest, are

- |                         |                    |                         |                    |
|-------------------------|--------------------|-------------------------|--------------------|
| <input type="radio"/> A | solid, liquid, gas | <input type="radio"/> B | gas, liquid, solid |
| <input type="radio"/> C | liquid, solid, gas | <input type="radio"/> D | gas, solid, liquid |

6. Which is an example of a noncrystalline solid?

- |                         |     |                         |       |
|-------------------------|-----|-------------------------|-------|
| <input type="radio"/> A | ice | <input type="radio"/> B | glass |
|-------------------------|-----|-------------------------|-------|

# Modern Chemistry Quiz Ch 9

**Michael Hoelscher**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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