Chapter 4 Earth Chemistry Review Answer Key

SECTION 1 MATTER

Review

- 1. first row: Electron second row, from left to right: +; -; none third row: in the nucleus
- Physical properties can be observed without chemical reactions. Chemical properties can be observed only when a substance reacts with another substance.
- The atom's mass number is 29. It is an atom of silicon. Silicon's atomic number is 14, so all atoms of silicon have 14 protons.
- Isotopes have the same number of protons, so their atomic numbers are equal. Isotopes have different numbers of neutrons, so their mass numbers are different.
- 5. They have the same number of valence electrons, and they have similar properties.

SECTION 2 COMBINATIONS OF ATOMS

Review

- second row, from left to right: compounds; mixtures third row, from left to right: covalent bonds; ionic bonds; heterogeneous; homogeneous
- The reactants are hydrogen (H_i) and oxygen (O_i). The product is water (H_iO). The equation is balanced because there are four hydrogen atoms and two oxygen atoms on each side of the equation.
- Possible answers: Coffee is an example of a homogeneous solution. Gravel is an example of a heterogeneous solution.

Chapter 5 Minerals Review Answer Key

SECTION 1 WHAT IS A MINERAL?

Review

- 1. Silicate minerals contain compounds of silicon and oxygen. Nonsilicate minerals do not.
- 2. cubes, hexagonal prisms, and irregular shapes
- Possible answer: Diamonds are minerals. They are solid, they form naturally, and they form crystals.
- 4. sulfides
- 5. Quartz is a silicate mineral, so quartz is made up of silicon-oxygen tetrahedrons. In a framework silicate such as quartz, each silicon-oxygen tetrahedron shares four oxygen atoms with other tetrahedrons. The tetrahedrons form a web, or framework. In a single-chain silicate such as pyroxene, each tetrahedron shares two oxygen atoms with other tetrahedrons. The tetrahedrons form long chains.
- 6. They do not contain silicon. All silicate minerals contain silicon and oxygen bonded together.

SECTION 2 IDENTIFYING MINERALS

Review

Review

- cleavage only: mineral breaks along smooth, flat surfaces cleavage and fracture: describes how a mineral breaks fracture only: mineral breaks along uneven surfaces
- If you scratched a piece of gold with a piece of talc, nothing would happen. The talc would not scratch the gold. Both the calcite and the quartz would scratch the gold.
- A mineral may have many different colors, but its streak is always the same, no matter what its color is.
- 4. double refraction

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