

# MOLE CONCEPT

## CLASS X<sup>th</sup>

1. Calculate the relative molecular masses (or molecular weights) of the following compounds :

- i) Copper sulphate crystals,  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ .
- ii) Ammonium sulphate,  $(\text{NH}_4)_2\text{SO}_4$ .
- iii) Cane sugar,  $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ .

(Given atomic masses Cu = 63.5, S = 32, O = 16, N = 14, C = 12)

2. An organic compound with vapour density = 94 contains. C = 12.67%, H = 2.13%, and Br = 85.11%. Find the molecular formula. [Atomic mass : C = 12, H = 1, Br = 80]

3. (i) The compound A has the following percentage composition by mass. Carbon = 26.7%, oxygen = 71.1%, hydrogen = 2.2%. Determine the Empirical formula of compound A [work to one decimal place] [H = 1, C = 12, O = 16].

ii) If the relative molecular mass of A is 90, what is the molecular formula of A?

(iii) The compound A is weak acid. What is meant by this statement?

4. Determine the empirical formula of the compound whose composition by mass is: 42% nitrogen, 48% oxygen and 9% hydrogen. [H = 1; N = 14; O = 16]

5. Determine the empirical formula of a compound containing 47.9% potassium, 5.5% beryllium and 46.6% fluorine by mass. (Atomic weight of Be = 9; F = 19; K = 39) Work to one decimal place.

6. The Empirical formula of a compound is  $\text{C}_2\text{H}_5$ . It has a vapour density of 29. Determine the relative molecular formula mass of the compound and hence its molecular formula.

7. The percentage composition of sodium phosphate as determined by analysis, is 42.1% sodium, 18.9% phosphorus and 39% oxygen. Find the empirical formula of the compound (work to two decimal places).

8. A gaseous organic compound contains 3.6 g of carbon and 0.8 g of hydrogen. The vapour density of this compound is 22. (i) Calculate the Empirical formula. (ii) Calculate the molecular formula of the compound.

# Mole Concept Chemistry Puzzles Answers

**Clifford Lane**



## **Mole Concept Chemistry Puzzles Answers:**

## Whispering the Techniques of Language: An Emotional Journey through **Mole Concept Chemistry Puzzles Answers**

In a digitally-driven earth wherever monitors reign great and quick connection drowns out the subtleties of language, the profound strategies and mental nuances concealed within phrases frequently get unheard. Yet, set within the pages of **Mole Concept Chemistry Puzzles Answers** a charming fictional treasure blinking with organic thoughts, lies an extraordinary journey waiting to be undertaken. Penned by a talented wordsmith, this wonderful opus attracts viewers on an introspective trip, delicately unraveling the veiled truths and profound influence resonating within the material of each word. Within the emotional depths of this moving review, we will embark upon a honest exploration of the book's core styles, dissect their interesting publishing style, and fail to the powerful resonance it evokes serious within the recesses of readers' hearts.

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