Lewis Structure Worksheet #2

(Key)

Draw Lewis Structures for each of the following molecules or ions:

1. Br:		2. BH ₃ H _B H
0 = 0		4. PS3. 24 :5 = P - S: + resonance
5. H ₂ CS /2 H		6. CHCls : 4: H - C - ::::
7. SF2 2.0 2.5 - 5 - F.		8. SHF
9. N.F. F. N-N-F.		10. XeO, Xe O:
11. N ₂ O ₃ (dinitrogen trioxide) $O = N - N = 0$	28e resonance	12. NO2 (nitrogen dioxide)* 17e
13. N ₂ O ₄ (dinitrogen tetraoxide)* N -N + reson	34 wance	14. N ₂ O ₅ (dinitrogen pentaoxide)* Hae

*Problems 11-14 are more challenging and may have more than one correct structure. Please note that if a molecule has an odd number of valence electrons, at least one atom in the molecule will NOT fulfill the octet rule.

Lewis Structure Worksheet Answer Key

Steve Lewis

Lewis Structure Worksheet Answer Key:

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