

7-3 Review and Reinforcement

Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

| Number | Prefix |
|--------|--------|
| 1 | mono |
| 2 | di |
| 3 | tri |
| 4 | tetra |
| 5 | penta |
| 6 | hexa |
| 7 | hepta |
| 8 | octa |
| 9 | nona |
| 10 | deca |

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation
anion

hydrates
anhydrates

electronegative
hydrogen ions

ionic
molecular

- Ionic compounds that absorb water into their solid structures form substances called **—hydrate**.
- When naming an ionic substance, the name of the **—cation** comes first.
- To name a binary molecular compound, add the suffix *-ide* to the more **—electronegative** of the two elements.
- By one definition, an acid is a molecular compound that dissolves in water to produce **—hydrogen ions**.
- The name of an acid is usually derived from the name of its characteristic **—anion**.
- Prefixes are often used when naming **molecular** compounds.

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Hal M. Saunders



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Medicinal Chemistry Norma K Dunlap, Donna M Huryn, 2025-11-28 This book provides a concise introduction to the medicinal chemistry of therapeutically active compounds Written in an accessible style it offers a brief overview of the drug development process from discovery to approved drug Emphasizing medicinal chemistry strategies the content is organized by target as well as therapeutic area drug classes It covers existing drugs and the discovery process that led to their development Aimed at students and professionals interested in drug discovery Medicinal Chemistry requires little prior knowledge making it ideal for classroom and training seminar use Written by experts in the field this outstanding new edition explains medicinal chemistry on

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American Chemical Review, 1885

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

Principles of Modern Chemistry David W. Oxtoby, H. P. Gillis, Norman H. Nachtrieb, 2002

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Table of Contents Modern Chemistry Chapter 7 Review Answers

1. Understanding the eBook Modern Chemistry Chapter 7 Review Answers
 - The Rise of Digital Reading Modern Chemistry Chapter 7 Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter 7 Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Chapter 7 Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter 7 Review Answers
 - Personalized Recommendations
 - Modern Chemistry Chapter 7 Review Answers User Reviews and Ratings
 - Modern Chemistry Chapter 7 Review Answers and Bestseller Lists
5. Accessing Modern Chemistry Chapter 7 Review Answers Free and Paid eBooks
 - Modern Chemistry Chapter 7 Review Answers Public Domain eBooks
 - Modern Chemistry Chapter 7 Review Answers eBook Subscription Services
 - Modern Chemistry Chapter 7 Review Answers Budget-Friendly Options
6. Navigating Modern Chemistry Chapter 7 Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Modern Chemistry Chapter 7 Review Answers Compatibility with Devices
 - Modern Chemistry Chapter 7 Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Chemistry Chapter 7 Review Answers
 - Highlighting and Note-Taking Modern Chemistry Chapter 7 Review Answers
 - Interactive Elements Modern Chemistry Chapter 7 Review Answers
8. Staying Engaged with Modern Chemistry Chapter 7 Review Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Modern Chemistry Chapter 7 Review Answers
- 9. Balancing eBooks and Physical Books Modern Chemistry Chapter 7 Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Chapter 7 Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Modern Chemistry Chapter 7 Review Answers
 - Setting Reading Goals Modern Chemistry Chapter 7 Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modern Chemistry Chapter 7 Review Answers
 - Fact-Checking eBook Content of Modern Chemistry Chapter 7 Review Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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