

Assessment

Chapter Test B**Teacher Notes and Answers****8** *Chemical Equations and Reactions***TEST B**

1. b
2. a
3. b
4. d
5. a
6. b
7. d
8. a
9. coefficient
10. liquid
11. reversible reaction
12. $A + X \rightarrow AX$
13. $AX \rightarrow A + X$
14. $A + BX \rightarrow AX + B$
15. $AX + BY \rightarrow AY + BX$
16. metal hydroxides
17. combustion reaction
18. electrolysis
19. activity series
20. chemical equation
21. conservation of mass
22. gaseous
23. heated
24. A word equation and a formula equation are both qualitative. Although they both show the reactants and products of a chemical reaction, they give no information about amounts of reactants or products. A chemical equation, however, does show the relative amounts of the reactants and products in a chemical reaction.
25. Three indications of a chemical reaction are the radiation of heat and light, the production of a gas, and the formation of a precipitate.
26. zinc sulfide + oxygen $\rightarrow$
zinc oxide + sulfur dioxide
27. sodium iodide + chlorine $\rightarrow$
sodium chloride + iodine
28. $Mg + O_2 \rightarrow MgO$
29. $CaCO_3 \rightarrow CaO + CO_2$
30. decomposition reaction
31. single-displacement reaction
32. synthesis reaction
33. double-displacement reaction
34. $2AgNO_3 + CuCl_2 \rightarrow 2AgCl + Cu(NO_3)_2$
35. $2PbO_2 \rightarrow 2PbO + O_2$
36. $Zn(OH)_2 + 2CH_3COOH \rightarrow$
 $Zn(CH_3COO)_2 + 2H_2O$
37. $Mg(s) + 2H_2O(g) \rightarrow Mg(OH)_2(aq) + H_2(g)$
38. no reaction
39. $Cl_2(g) + MgBr_2(aq) \rightarrow MgCl_2(aq) + Br_2(l)$
40. $Zn(s) + 2HCl(aq) \rightarrow ZnCl_2(aq) + H_2(g)$
41. no reaction
42. $H_2(s) + CuCl_2(aq) \rightarrow H_2Cl_2(aq) + Cu(s)$
43. no reaction
44. $Mg(s) + Co(NO_3)_2(aq) \rightarrow$
 $Mg(NO_3)_2(aq) + Co(s)$

Modern Chemistry Chapter 8 Test Answer Key

Joseph William Mellor



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Elementary Modern Chemistry Wilhelm Ostwald, Harry Wheeler Morse, 1909 **Laboratory Experiments to Accompany "Modern Chemistry,"** Fredus Nelson Peters, 1902 **Modern Electrochemistry 2B** John O'M. Bockris, Amulya K.N. Reddy, 2007-05-08 This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered The lectures were recorded and written up by Dr Klaus Muller as a 293 page manuscript At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met Corrosion is recognized as having an electrochemical basis The synthesis of nylon now contains an important electrochemical stage Some central biological mechanisms have been shown to take place by means of electrochemical reactions A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States Essentials of Modern Chemistry Charles Elwood Dull, 1918 Principles of Modern Organic Chemistry James Cason, 1966 Modern inorganic chemistry Joseph William Mellor, 1918 **Modern Chemistry** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Books in Print Supplement*, 2002 *Modern Chemical Technology* Chemical Technician Curriculum Project. Writing Team, 1972 Edexcel A Level Chemistry Student Book 1 Andrew Hunt, Graham Curtis, Graham Hill, 2015-07-17 Exam Board Edexcel Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Endorsed by Edexcel Develop and assess your students knowledge and mathematical skills throughout A Level with worked examples practical assessment guidance and differentiated end of topic questions with this Edexcel Year 1 student book Identifies the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Year 1 Student Book Provides support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Mathematical skills are integrated throughout with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offers plenty of practice with Test Yourself Questions to help students assess their understanding and measure progress Encourages further reading and study with short passages of extension material Develops understanding with free online access to Test yourself Answers and an Extended Glossary Edexcel A level Chemistry Year 1 Student Book includes AS

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