

Chapter 12 Review 2

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- 1 A mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together is called a(n)
 - a. dipole.
 - b. Lewis structure.
 - c. chemical bond.
 - d. London force.
- 2 The electrons involved in the formation of a chemical bond are called
 - a. dipoles.
 - b. s electrons.
 - c. Lewis electrons.
 - d. valence electrons.
- 3 A chemical bond resulting from the electrostatic attraction between positive and negative ions is called a(n)
 - a. covalent bond.
 - b. ionic bond.
 - c. charged bond.
 - d. dipole bond.
- 4 If two covalently bonded atoms are identical, the bond is
 - a. nonpolar covalent.
 - b. polar covalent.
 - c. nonionic.
 - d. coordinate covalent.
- 5 If the atoms that share electrons have an unequal attraction for the electrons, the bond is called
 - a. nonpolar.
 - b. polar.
 - c. ionic.
 - d. dipolar.
- 6 A covalent bond results when _____ are shared.
 - a. ions
 - b. Lewis structures
 - c. electrons
 - d. dipoles
- 7 Nonpolar covalent bonds are not common because
 - a. one atom usually attracts electrons more strongly than the other.
 - b. ions always form when atoms join.
 - c. the electrons usually remain equally distant from both atoms.
 - d. dipoles are rare in nature.
- 8 The greater the electronegativity difference between two bonded atoms, the greater the percentage of
 - a. ionic character.
 - b. covalent character.
 - c. metallic character.
 - d. electron sharing.
- 9 The pair of elements that forms a bond with the least ionic character is
 - a. Na and Cl.
 - b. H and Cl.
 - c. O and Cl.
 - d. Br and Cl.
10. In which of these compounds is the bond between the atoms NOT a nonpolar covalent bond?
 - a. Cl_2
 - b. H_2
 - c. HCl
 - d. O_2

Modern Chemistry Chapter Practice Test

Karin Nielsen-Saines



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medium term What has changed most significantly in the past few years is the time available for identifying new chemical entities Because of the high costs of drug discovery projects pressure for maximum success in the shortest possible time is higher than ever In addition the multidisciplinary character of the field is much more pronounced today than it used to be As a consequence researchers and project managers in the pharmaceutical industry should have a solid knowledge of the more important methods available to drug discovery because it is the rapidly and intelligently combined use of these which will determine the success or failure of preclinical projects

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