

Chemistry Worksheets Class 11 on Chapter 11 The p-Block Elements- Set 5

Q-1: Which of the following Group 14 elements has the lowest atomic volume?

- a) St
- b) C
- c) Sm
- d) Ge

Q-2: Which of the following is an amorphous form of carbon?

- a) Graphite
- b) Lamp-black
- c) Diamond
- d) All of the above

Q-3: Which of the following types of graphite has an ABAB-type layer arrangement?

- α-graphite
- b) β-graphite
- c) Both a) and b)
- d) None of the above

Q-4: Conc. HNO, is stored in containers made up of

- a) Zinc
- b) Aluminium
- c) Tim
- d) Copper

Q-5: Which of the following is not a protonic acid?

- a) B(OH),
- b) PO(OH),
- c) SO(OH)₂
- d) $SO_2(OH)_2$

Q-6: Standard electrode potential values, E^o for Al³*/Al is -1.66 V and +1.26 V for Tl³*/Tl. Predict the creation of the M³* ion in solution and contrast the electropositive properties of the two metals.

Q-7: Boron is unable to form BF. 3- ion. Explain.

Q-8: Why is boric acid regarded as a weak acid?

Modern Chemistry Worksheet Answers Chapter 11

DP Hallahan

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