## Chemistry: Molarity

Directions: Using these formulas, solve the following problems



molarity = mol solute / L solution 1 amu = 1 gram / mol

M.W. = MIW.

Problem 1: Calculate the molarity of a 2.21 L solution made with 317 g of calcium chloride Answer Problem 2. How many grams of NaCl would be required to produce a 6.1 M solution with a volume of 0.64 L? Answer. Problem 3. How many grams of NaCH would be required to produce a 5.36 M solution. with a volume of 1.37 L7 Answer. Problem 4. How many grams of KCI would you need to weigh out to prepare 0.416 L of a 3.37 M KCI salution? Answer. Problem 5: Calculate the molarity of a 6.68 mt, solution of 6.89 a NaCl. Answer \_\_\_\_ Problem 6: Calculate the volume in ml. if you have 18.9 g of Na<sub>2</sub>CO<sub>2</sub> from a 0.112 M solution Answer Problem 7: Calculate the volume in mL if you have 28.1 g LiOH from a 7.11 M solution. Arthwest Problem B. Calculate the final concentration if water is added to 0.789 L of a 3.68 M HCI solution to give a volume of 0.387 L. Answer Problem 9: Calculate the final concentration if a 8.63 mL of a 1.12 M HCI solution is diluted with water to 2.54 mL. Answer. Problem 10: Caliculate the final concentration if water is added to 0.235 L of a 6.11 M KBr solution to give a volume of 5.79 L. Answer.

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