# Mole Ratios

MATALIA How can the coefficients in a chemical equation be interpreted?

Why?

A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles. However, in real life the reaction represented by an equation occurs an unimaginable number of times. Short of writing very large numbers (102) or larger) in front of each chemical in the equation, how can we interpret chemical equations so that they more realistically represent what is happening in real life? In this activity you will explore the different ways a chemical reaction can be interpreted.

## Model 1 – A Chemical Reaction

 $N_{s}(g) + 3H_{s}(g) \rightarrow 2NH_{s}(g)$ 

- Consider the reaction in Model 1.
  - a. What are the coefficients for each of the following substances in the reaction?

4 N.

3, H,

JINH.

6. Draw particle models below to illustrate the reaction in Model 1.

00+ = --- 4 of

- Consider each situation below as it relates to the reaction in Model 1.
  - Calculate the amount of reactants consumed and products made.
  - & Record the ratio of N, to H, to NH, Reduce the ratio to the lowest whole numbers possible.

	N <sub>z</sub> Consumed	H <sub>1</sub> Consumed	NH <sub>3</sub> Produced	Ratio N.:H.:NH, (reduced)
For a single reaction, how many molecules of each substance would be consumed or produced?	1	3	2	1:3:2
If the reaction occurred one hun- dred times, how many molecules would be consumed or produced?	100	300	200	1:3:2
If the reaction occurred 538 times, how many molecules would be consumed or produced?	53%	1614	1076	1:3:4

# **Mole Ratios Worksheet Model 1 A**

**Theodore B. Selover** 

#### **Mole Ratios Worksheet Model 1 A:**

Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 Thermophysical Properties for Industrial Process Design Theodore B. Selover, 1994 his volume presents papers from the 1992 Symposium on Thermophysical Properties for Industrial Process Design Papers in Part I deal with new experimental techniques to provide data that are either difficult to measure by a conventional method or new data sets leading to the development of estimation or correlation of data new property estimation methods software for chemical equilibrium calculation and use of data in process design Papers in Part II cover novel experiences in industrial applications of existing models and methods for property estimation and phase equilibrium calculations Also included are discussions of software for database management and for molecule design by property The industrial processes covered in these papers involve a range of chemical systems including non electrolytes surfactants etc A detailed key word index is also provided Chemical Engineering Education Introduction to Software for Chemical Engineers Mariano Martín Martín, 2025-03-24 The field of chemical engineering and its link to computer science is in constant evolution and engineers have an ever growing variety of tools at their disposal to tackle everyday problems Introduction to Software for Chemical Engineers Third Edition provides a quick guide to the use of various computer packages for chemical engineering applications. It covers a range of software applications including Excel and general mathematical packages such as MATLAB MathCAD R and Python Coverage also extends to process simulators such as CHEMCAD HYSYS and Aspen equation based modeling languages such as gPROMS optimization software such as GAMS AIMS and Julia and specialized software like CFD or DEM codes The different packages are introduced and applied to solve typical problems in fluid mechanics heat and mass transfer mass and energy balances unit operations reactor engineering and process and equipment design and control This new edition is updated throughout to reflect software updates and new packages It emphasizes the addition of SimaPro due to the importance of life cycle assessment as well as general statistics software SPSS and Minitab that readers can use to analyze lab data The book also includes new chapters on flowsheeting drawing process control and LOOP Pro as well as updates to include Pyomo as an optimization platform reflecting current trends The text offers a global idea of the capabilities of the software used in the chemical engineering field and provides examples for solving real world problems Written by leading experts this handbook is a must have reference for chemical engineers looking to grow in their careers through the use of new and improving computer software Its user friendly approach to simulation and optimization as well as its example based presentation of the software makes it a perfect teaching tool for both undergraduate and graduate level readers **Working with Chemistry** Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day

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Transactions of the American Nuclear Society ,1987

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