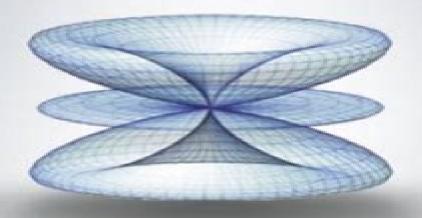
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Orbitals in Chemistry Victor M. S. Gil,2000-08-10 This text presents a unified and up to date discussion of the role of atomic and molecular orbitals in chemistry from the quantum mechanical foundations to the recent developments and applications. The discussion is mainly qualitative largely based on symmetry arguments. It is felt that a sound mastering of the concepts and qualitative interpretations is needed especially when students are becoming more and more familiar with numerical calculations based on atomic and molecular orbitals. The text is mathematically less demanding than most traditional quantum chemistry books but still retains clarity and rigour. The physical insight is maximized and abundant illustrations are used. The relationships between the more formal quantum mechanical formalisms and the traditional chemical descriptions of chemical bonding are critically established. This book is of primary interest to undergraduate chemistry students and others taking courses of which chemistry is a significant part. Molecular Quantum Mechanics. P. W. Atkins, 1970. Reliability Physics and Engineering. J. W. McPherson, 2013-06-03. Reliability Physics and Engineering provides critically important information for designing and building reliable cost effective products. The textbook contains

numerous example problems with solutions Included at the end of each chapter are exercise problems and answers Reliability Physics and Engineering is a useful resource for students engineers and materials scientists Quanta, Matter and Change: A Molecular Approach to Physical Change Peter Atkins, Julio de Paula, Ron Friedman, 2008-11-15 Beginning with quantum mechanics introducing statistical mechanics and progressing through to thermodynamics this new text for the two semester physical chemistry course features a wealth of new applications and insights as well as new Mathematical Background inter chapters to help students review key quantitative concepts This is a splendid book True to the authors philosophy as outlined in the preface it approaches physical chemistry by first developing the quantum theory of molecular electronic structure then by statistical arguments moves into thermodynamics and thence to kinetics Peter Taylor Review in Chemistry World Royal Society of Chemistry July 31 2009 Molecular Quantum Mechanics/ Solutions Manual Peter William Atkins, 1983 Quantum Mechanics Lanbrecht Piet Kok, J. Visser, 1987 Chemistry, Molecular Quantum Mechanics P. W. Atkins, R. S. Friedman, 1997

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