# Rate of Reachon:

Rate = Change in Conc or Pressure. Time taken

Wut - and L' T" or olm T"

\* ROR conneverbe megalive (-ve)

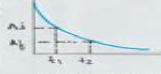
# Rate of Reachen.

Average Rate

$$Rode = \frac{\Delta(A)}{\Delta t} = \frac{\Delta B}{\Delta t}$$

Instantoneous Rate

$$\mathsf{Rate} = -\frac{\mathsf{d}[\mathsf{A})}{\mathsf{d} \mathsf{r}} = -\frac{\mathsf{d}[\mathsf{B}]}{\mathsf{d} \mathsf{t}}$$





$$\underset{\text{ang}}{\text{R}} \left( \frac{A_{b} - A_{c}}{t_{k} - t_{l}} \right)$$

Rade = 
$$\frac{-4DN_0}{dt}$$
 =  $\frac{-4DN_0}{3dt}$  =  $\frac{1}{3}\frac{dDNN_0}{dt}$ 

Rate of Disappearant = 
$$-\frac{dDND}{dt} = -\frac{dDND}{dt}$$

ROD & ROA does not include the St. Coeff. aA + bB -> aC+dD

$$ROR = \frac{-1}{a} \frac{d(A)}{dt} = \frac{-1}{b} \frac{d(B)}{dt} = \frac{1}{c} \frac{d(C)}{dt} = \frac{1}{d} \frac{d(D)}{dt}$$

~ Differential Rate Law

9. The scale of the occasion: 
$$2N_2O_6 \rightarrow 4No_2 + O_2$$
Can be exemitten in three ways  $\rightarrow$ 

$$-d [N_2O_2] = K [N_2O_3] ; d [No_2] = k' (N_2O_3)$$

$$dt = K''[N_2O_3]$$

$$RoR = \frac{1}{2} \frac{4[w_{c}v_{d}]}{dt} = \frac{4[v_{d}]}{dt} = \frac{4[v_{d}]}{dt}$$

$$= \frac{K[w_{c}v_{d}]}{2} - \frac{K'[w_{c}v_{d}]}{4} = \frac{K'}{4} \frac{[w_{c}v_{d}]}{4}$$

$$RoR = K' = 2K \quad K'' = K$$

# Moleculomby :

A Can be integer (1,2 or 8) but it cannot be zero, megative or prachonol.

A For elementary sceachon: Molecularity is equal to Ms ander

A Maximum value of molecularity is 3. (en ender)

A For complex own moleculority of each Sterp of mechanism is defined separately. · moleculously commot be deferred for whole complet Reachon - or once -

A Molecular ply of a scenchion may as may not be equal to order of sucoction.

A Molecularity can not be determined by balanced chemical vixin and can only be determed from Dimple reaching

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**Charles G. Hill** 

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