

## LIMITING REACTANTS

1. What mass of  $\text{CS}_2$  is produced when 17.5 g of C are reacted with 39.5 g of  $\text{SO}_2$  according to the equation below?



What mass of the excess reactant will be left over?

2. What mass of NO is produced when 87.0 g of Cu are reacted with 225 g of  $\text{HNO}_3$  according to the equation?



What mass of the excess reactant will be left over?

3. What mass of  $\text{P}_4$  is produced when 41.5 g of  $\text{Ca}_3(\text{PO}_4)_2$ , 26.5 g of  $\text{SiO}_2$  and 7.80 g of C are reacted according to the equation



How many grams of each excess reactant will remain unreacted?

# Limiting Reactants And Percent Yield 9 3 Section Review

**Marian DeWane, Heather Hattori**



## **Limiting Reactants And Percent Yield 9 3 Section Review:**

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

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