

Can You Find the Mole Ratio?

In this experiment, baking soda (NaHCO_3) reacts with hydrochloric acid (HCl) to produce sodium chloride, carbon dioxide, and water. You will determine the moles of reactant used and moles of product produced using both the lab data and stoichiometry and compare the two.

Data Table:

mass of test tube and baking soda	19.46 g
mass of empty test tube	19.21 g
mass of baking soda	0.25 g
mass of test tube and sodium chloride	19.39 g
mass of empty test tube	19.21 g
mass of sodium chloride	0.18 g

Conclusion Questions:

1. Calculate the number of moles of baking soda used in the lab.

$$0.25 \text{ g NaHCO}_3 \left(\frac{\text{mole}}{84.01 \text{ g}} \right) = 0.00297 \text{ moles of NaHCO}_3$$

2. Calculate the number of moles of sodium chloride produced in the lab.

$$0.18 \text{ g of NaCl} \left(\frac{\text{mole}}{58.44} \right) = 0.0031 \text{ moles of NaCl}$$

3. What is the experimental mole ratio of baking soda (NaHCO_3) to sodium chloride (NaCl)?

$$1:1$$

4. Write a balanced equation for the reaction that took place in the experiment.



5. According to the balanced equation, what is the theoretical mole ratio of baking soda to sodium chloride?

$$1:1$$

6. Was the experimental mole ratio exactly the same as the theoretical mole ratio? If not, give some possible reasons.

Yes ~ if it wasn't some reasons

side reactions
measurement

transfer
contamination

Lab Stoichiometry Datasheet Key

C. Wayne Smith

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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... **Catalog** National Institute of Standards and Technology (U.S.),1991 *Archaeological Conservation Using Polymers* C.
Wayne Smith,2003 Annotation Over the years archaeologists have developed a number of techniques for conserving
historical artifacts for future generations Along with these techniques researchers have developed a series of ethical
principles for treating materials in a way that allows them to be not only observed and analyzed for the present but also re
studied in the future Conservation techniques used up to now however have provided artifacts only a limited lifespan and in
some cases they do not work well with waterlogged materials Within the past few years archaeological chemistry and
concerns of longevity testing have become central issues in the development of conservation treatment strategies Working
with Dow Corning Corporation Texas A M s Archaeological Preservation Research Lab APRL and the Conservation Research
Lab CRL Smith and his colleagues in AS M s Nautical Archaeology Program set out to develop a series of chemistries and
techniques that would provide successful and affordable treatment strategies for organic materials In this ground breaking
description of the processes and materials that were developed Smith explains these techniques in ways that will allow
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Variance in equipment relative humidity laboratory layout intended results and level of expertise all affect researchers ability
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wishing to incorporate some or all of them will find the step by step instructions for doing so BOOK JACKET Title Summary
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