

Ethyl alcohol	$C_2H_5OH$	46.068	l	-277.63	-1366.91
			g	-235.31	-1409.25
Ethyl benzene	$C_6H_5 \cdot C_2H_5$	106.16	l	-12.46	-4564.87
			g	29.79	-4607.13
Ethyl chloride	$C_2H_5Cl$	64.52	g	-105	
Ethylene	$C_2H_4$	28.052	g	52.283	-1410.99
Ethylene chloride	$C_2H_3Cl$	62.50	g	31.38	-1271.5
3-Ethyl hexane	$C_8H_{18}$	114.22	l	-250.5	-5470.12
			g	-210.9	-5509.78
Ferric chloride	$FeCl_3$		c	-403.34	
Ferric oxide	$Fe_2O_3$	159.70	c	-822.156	
Ferric sulfide	$FeS_2$	see Iron sulfide	see Iron sulfide		
Ferrosoferric oxide	$Fe_3O_4$	231.55	c	-1116.7	
Ferrous chloride	$FeCl_2$		c	-342.67	-303.76
Ferrous oxide	$FeO$	71.85	c	-267	
Ferrous sulfide	$FeS$	87.92	c	-95.06	
Formaldehyde	$H_2CO$	30.026	g	-115.89	-563.46
n-Heptane	$C_7H_{16}$	100.20	l	-224.4	-4816.91
			g	-187.8	-4853.48
n-Hexane	$C_6H_{14}$	86.17	l	-198.8	-4163.1
			g	-167.2	-4194.753
Hydrogen	$H_2$	2.016	g	0	-285.84
Hydrogen bromide	$HBr$	80.924	g	-36.23	
Hydrogen chloride	$HCl$	36.465	g	-92.311	
Hydrogen cyanide	$HCN$	27.026	g	130.54	
Hydrogen sulfide	$H_2S$	34.082	g	-20.15	-562.589
Iron sulfide	$FeS_2$	119.98	c	-177.9	
Lead oxide	$PbO$	223.21	c	-219.2	
Magnesium chloride	$MgCl_2$	95.23	c	-641.83	
Magnesium hydroxide	$Mg(OH)_2$	58.34	c	-924.66	
Magnesium oxide	$MgO$	40.32	c	-601.83	

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**Dean Smith, John Kilgo, Sally Jenkins**



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