



BUSINESS STUDIES
GRADE 12
TERM ONE
CHAPTER 3
NOTES ON ETHICS AND PROFESSIONALISM
2019

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web the atomic mass of a single atom is simply its total mass and is typically expressed in atomic mass units or amu by definition an atom of carbon with six neutrons carbon 12 has an atomic mass of 12 amu other atoms don t generally have round number atomic masses for reasons that are a little beyond the scope of this article

chapter 1 7 the mole and molar mass chemistry libretexts - Jul 02 2022

web jul 29 2021 as you learned the mass number is the sum of the numbers of protons and neutrons present in the nucleus of an atom the mass number is an integer that is approximately equal to the numerical value of the atomic mass although the mass number is unitless it is assigned units called atomic mass units amu because a molecule or a

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web atomic structure google classroom what three particles make up an atom choose 1 answer protons neutrons and photons a protons neutrons and photons positrons neutrons and electrons b positrons neutrons and electrons protons electrons and positrons c protons electrons and positrons protons neutrons and electrons d