

Instructions: Use the appropriate equation that best answers each question and give your answer using the appropriate units for the quantity being provided. The answers may vary by rounded values. The tolerance is ± 0.01 when noted.

Answer choice: When told to give a value that pertains to the "molar mass of a substance", the answer is to give the molar mass of the substance. When told to give a value that pertains to the "molar mass of a substance", the answer is to give the molar mass of the substance. When told to give a value that pertains to the "molar mass of a substance", the answer is to give the molar mass of the substance.

1. Chemical equation: $\text{C}_2\text{H}_5\text{OH} + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$. How many moles of O_2 are needed to react with 1.00 mole of $\text{C}_2\text{H}_5\text{OH}$?
 A) 1.00 mol ☐ B) 1.50 mol ☒ C) 3.00 mol ☐ D) 4.50 mol
2. Which of the following is a chemical equation? ☐ A) $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$ ☐ B) $\text{H}_2\text{O} + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$ ☒ C) $\text{H}_2\text{O} + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$ ☐ D) $\text{H}_2\text{O} + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$
3. Which of the following is a chemical equation? ☐ A) $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$ ☐ B) $\text{H}_2\text{O} + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$ ☒ C) $\text{H}_2\text{O} + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$ ☐ D) $\text{H}_2\text{O} + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$
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