#### SHORT ANSWER STUDY GUIDE QUESTIONS - The Giver

#### Chapters 1-2

- What did the word "frightened" mean, according to Jonas?
- 2. What were Jonas and the other children taught to be careful about?
- How did Jonas decide he felt? What was causing this feeling?
- 4. What evening ritual did the family perform after dinner?
- 5. What were the two occasions when release was not punishment?
- 6. What did Father want to do about the new child, and why?
- 7. What fascinated Jonas about his father?
- 8. Why was the Ceremony of Twelve so important?
- Based on what you have read so far, how would you describe Jonas' community? Be sure to use the strongest evidence
- How are families formed in Jonas's community? In what ways is this similar and different to our world? Be sure to include evidence from the novel in your answer.

#### Chapters 3-5

- 1. What was unusual about Jonas and the new child?
- Describe the Assignment of Birthmother. What did Lily's parents say about it?
- 3. What happened to the apple while Jonas was playing with it?
- 4. Describe the Celebration of Release of Roberto,
- Did Larissa know exactly where Roberto or anyone else went when they were released?
- 6. What was Jonas's dream about? What did his mother and father say about it?
- What is the role of the birthmother? Why is this considered a job with "very little honor"? Be sure to use passages from the text to support your answer. (required)
- 8. Lily and Jonas face "chastisement" in this chapter. For what reasons are they chastised? Why might the community have the rules that they violate?
- How are the old treated in the community? Explain how this is similar and different from our society? Include at least two brief excerpts from the chapter to support your stance.
- 10. How does the pill for the stirrings help maintain Jonas' community as a utopia? What problems from our society might it solve?

#### Chapters 6-7

- Describe the jacket that the Fours, Fives, and Sixes wore, and the reason it was designed the way it was. Also describe the jacket the Sevens wore, and what it symbolized.
- 2. Describe the Ceremony. Tell what happened at the Naming and the other age levels.
- 3. What Assignment did Asher get?
- 4. What happened when Jonas's number should have been called? What did Jonas think?

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**Tobias Bleicker** 

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thermal energy mcgraw hill education - Jun 15 2023

web physical science chapter 6 thermal energy thermal energy your results the correct answer for each question is indicated by a 1 thermal energy is the of the particles in a material is the thermal energy that flows from a material with a higher temperature to one with a lower temperature

# chapter 16 thermal energy and heat henry county schools - Feb 28 2022

web thermal energy depends on the mass temperature and phase solid liquid or gas of an object thermal expansion occurs when particles of matter move farther apart as temperature increases thermal expansion is an increase in the volume of a material due to a temperature increase

what is heat lesson teachengineering - Jan 30 2022

web oct 31 2021 explain that heat is the flow of energy from hot materials to cold materials describe that molecules in a material begin to vibrate or move more quickly when the material is heated identify conduction as heat transfer within and between solids identify convection as heat transfer involving gases or liquids

physical science thermal energy chapter 5 worksheet quizlet - May 14 2023

web flashcards learn test match created by allisondayhoward17 terms in this set 12 temperature the increases when the average kinetic energy of the particles in a material increases radiation thermal energy transfer that does not require matter is heat thermal energy that flows from a higher to a lower temperature is

# chapter 5 thermal energy houston independent school district - Apr 01 2022

web how does thermal energy change the thermal energy of an object changes when heat flows into or out of the object you can use the following equation to calculate the change in thermal energy change in thermal energy j mass kg change in temperature c specific heat j kg c q m t f t i c