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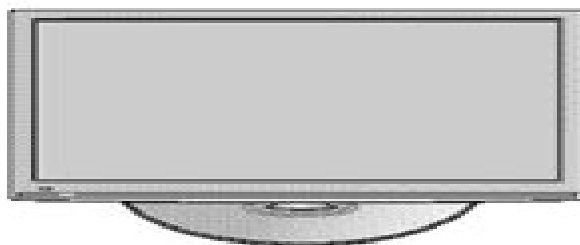
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c o c l 2 n f 3 n f 3 q 3

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as bf 3 trigonal planar ch 4 tetrahedral pf 5 trigonal bipyramidal and sf 6

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