C3 - STRUCTURE AND BONDING 3.5 Covalent bond

Aiming for Grade 4

Aiming for Grade 6

Aiming

Ionic bonds form between metals and non-metals, Covalent bonds form between non-metals. Covalent bonds form. when atoms share pairs of electrons. These bonds are very strong.

Share, Metals, strong, non-metals, electrons You can use the words more than once

Describe how Hydrogen and Chlorine form a covalent bond to form Hydrochloric acid.

To gain full outer shells Hydrogen needs to gain an electron and chlorine needs to lose one. Hydrogen shares its outer shell electron with Ollorine and Ollorine shares one of its electrons.

What is a double covalent bond? Two shared pairs of electrons

Draw ball and stick on the following covalent

Three hydrogen atoms one phosphorus (phosp

Carbon diaxide



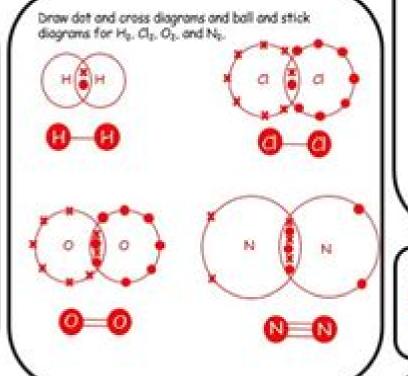
Ethene (2 carbon with hydnogen





How do double and trial properties? A double or triple bond the band making the me

Name these small molecules and state their formulae. Hydrochloric Acid (HCI) Water (HJ0) Methone (CHJ) Ammonia (NH₂)



Assessment

I can draw dot an-

Assessment

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