

Report Sheet – Experiment 8- Determination of an Empirical Formula

Part A – Determination of the empirical formula for magnesium chloride
Show calculations for all items marked with an asterisk (*).

1. Mass of clean, empty crucible	92.526
2. Mass of polished Mg strip	0.146 gm
3. Mass of crucible and dry magnesium chloride	93.098 gm
4. Mass of magnesium chloride*	total mass - mass empty crucible $93.098 - 92.526 = 0.572 \text{ gm}$
5. Mass of chlorine used to form the magnesium chloride*	mass of MgCl_2 - mass of Mg $0.572 - 0.146 = 0.426 \text{ gm}$
6. Moles of Mg used*	mass / molar mass $0.146 / 24 = 0.006$
7. Moles of Cl used*	$0.426 / 35.5 = 0.0012$
8. Ratio of moles of Mg to moles of Cl*	$0.006 / 0.0012 = 1 / 2$
9. Empirical formula of magnesium chloride	MgCl_2

Lab Determining An Empirical Formula Mgo Answers

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Nordulf**



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Smoot,Smoot,Richard G. Smith,Jack Price,1998 **Experiments in Chemistry** Frank R. Milio,Clyde R. Metz,W. G. Nordulf,1991-03 **I/M&w/Tsts Intro Chem** Victor S Krimsley,Darold E Skerritt,Beverly B Harrison,1986 Laboratory Experiments for Basic Chemistry Alan Sherman,Sherman,1988 **Introductory Chemistry** Charles H. Corwin,2005 For one semester courses in Basic Chemistry Introduction to Chemistry and Preparatory Chemistry and the first term of Allied Health Chemistry This text is carefully crafted to help students learn chemical skills and concepts more effectively Corwin covers math and problem solving early in the text he builds student confidence and skills through innovative problem solving pedagogy and technology formulated to meet student needs **Students Study Guide** Fredeen,Pearson,2003-07

Instructor's Manual with Test Bank [for] Basic Concepts of Chemistry, Fourth Edition Alan Sherman,1988

Teachers Manual to Accompany Laboratory for General Chemistry Beran,1986-05 *U.S. Government Research Reports* ,1964 **Instructor's Guide** Alan Sherman,Leonard Russikoff,Sharon Sherman,1976 *Engineering and Mining Journal* ,1900 *Journal of Industrial and Engineering Chemistry* ,1919 **The Journal of Industrial and Engineering Chemistry** ,1919

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