

Mole Ratios

How can the coefficients in a chemical equation be interpreted?

NATALIA
LACAYO
P. 2

Why?

A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles. However, in real life the reaction represented by an equation occurs an unimaginable number of times. Short of writing very large numbers (10^{23} or larger) in front of each chemical in the equation, how can we interpret chemical equations so that they more realistically represent what is happening in real life? In this activity you will explore the different ways a chemical reaction can be interpreted.

Model 1 – A Chemical Reaction



1. Consider the reaction in Model 1.

a. What are the coefficients for each of the following substances in the reaction?

1 N_2

3 H_2

2 NH_3

b. Draw particle models below to illustrate the reaction in Model 1.



2. Consider each situation below as it relates to the reaction in Model 1.

a. Calculate the amount of reactants consumed and products made.

b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole numbers possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$ (reduced)
For a single reaction, how many molecules of each substance would be consumed or produced?	1	3	2	1:3:2
If the reaction occurred one hundred times, how many molecules would be consumed or produced?	100	300	200	1:3:2
If the reaction occurred 538 times, how many molecules would be consumed or produced?	538	1614	1076	1:3:2

Mole Ratios Worksheet Model 1 A Chemical Reaction

Albert A Gayle



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Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 Chemical, Biochemical, and Engineering Thermodynamics Stanley I. Sandler, 2017-04-24 In this newly revised 5th Edition of Chemical and Engineering Thermodynamics Sandler presents a modern applied approach to chemical thermodynamics and provides sufficient detail to develop a solid understanding of the key principles in the field The text confronts current information on environmental and safety issues and how chemical engineering principles apply in biochemical engineering bio technology polymers and solid state processing This book is appropriate for the undergraduate and graduate level courses *Introduction to Software for Chemical Engineers* Mariano Martín Martín, 2025-03-24 The field of chemical engineering and its link to computer science is in constant evolution and engineers have an ever growing variety of tools at their disposal to tackle everyday problems Introduction to Software for Chemical Engineers Third Edition provides a quick guide to the use of various computer packages for chemical engineering applications It covers a range of software applications including Excel and general mathematical packages such as MATLAB MathCAD R and Python Coverage also extends to process simulators such as CHEMCAD HYSYS and Aspen equation based modeling languages such as gPROMS optimization software such as GAMS AIMS and Julia and specialized software like CFD or DEM codes The different packages are introduced and applied to solve typical problems in fluid mechanics heat and mass transfer mass and energy balances unit operations reactor engineering and process and equipment design and control This new edition is updated throughout to reflect software updates and new packages It emphasizes the addition of SimaPro due to the importance of life cycle assessment as well as general statistics software SPSS and Minitab that readers can use to analyze lab data The book also includes new chapters on flowsheeting drawing process control and LOOP Pro as well as updates to include Pyomo as an optimization platform reflecting current trends The text offers a global idea of the capabilities of the software used in the chemical engineering field and provides examples for solving real world problems Written by leading experts this handbook is a must have reference for chemical engineers looking to grow in their careers through the use of new and improving computer software Its user friendly approach to simulation and optimization as well as its example based presentation of the software makes it a perfect teaching tool for both undergraduate and graduate level readers **How to Solve Word Problems in Chemistry** David E Goldberg, 2001-06-26 In addition to having to master a vast number of difficult concepts and lab procedures high school chemistry students must also learn with little or no coaching from their teachers how to solve tough word problems Picking up where standard chemistry texts leave off How to Solve Word Problems in Chemistry takes the fear and frustration out of chemistry word problems by providing students with easy to follow procedures for solving problems in everything from radioactive half life to oxidation reduction reactions Addison-Wesley Chemistry , 1990 Chemical Engineering Education , 2004 **Cambridge IGCSE Chemistry Coursebook with CD-ROM** Richard Harwood, Ian Lodge, 2014-07-31

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