

- (b) Caesium-133 is a stable isotope of the element caesium, but caesium-135 is radioactive.

A nucleus of caesium-133 contains 78 neutrons and a nucleus of caesium-135 contains 80 neutrons.

Put **one** tick in each row of the table to indicate how the number of particles in a neutral atom of caesium-133 compares with the number of particles in a neutral atom of caesium-135.

The first row has been completed already:

	particles in caesium-133				
	2 more than caesium-135	1 more than caesium-135	equal to caesium-135	1 fewer than caesium-135	2 fewer than caesium-135
number of neutrons					✓
number of protons					
number of nucleons					
number of electrons					

[2]

[Total: 6]

Mark Scheme Of 33 M J 14

Minjie Lin



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