

Mark scheme

Chemistry Past Paper Questions 1

SECTION B: CHEMISTRY

Item	Expected answers	Additional guidance	Marks
4 (a)	5		1
4 (b)	An electron has a much smaller mass than a proton.		1
4 (c)	$(20 \times 10) + (80 \times 11)/100$ (1) = 10.8 (1)	$(200 + 80)/100$ (1) or $1080/100$ (1)	2
4 (d)	2 or 8 but not 3 for 1 mark. 2, 8, 3 for 2 marks.		2
Total mark			6

Item	Expected answers	Additional guidance	Marks
5 (a)	Any one of the following: It is on the right hand side of the periodic table (1) It is an inert gas/rare gas/noble gas (1)		1
5 (b)	Any two of the following: They have similar reactions to each other (1) They have similar properties to each other (1) They all have 1 electron in their outer shell (1)	Do not accept 'they are all alkali metals'	2
Total mark			3

Mark Scheme Chemistry 97june November2013 Paper 4

Thomas Griffiths



Mark Scheme Chemistry 97june November2013 Paper 4:

The book delves into Mark Scheme Chemistry 97june November2013 Paper 4. Mark Scheme Chemistry 97june November2013 Paper 4 is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Mark Scheme Chemistry 97june November2013 Paper 4, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Mark Scheme Chemistry 97june November2013 Paper 4
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 - Chapter 3: Mark Scheme Chemistry 97june November2013 Paper 4 in Everyday Life
 - Chapter 4: Mark Scheme Chemistry 97june November2013 Paper 4 in Specific Contexts
 - Chapter 5: Conclusion
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