

This resource includes exam questions from both Chemistry A and Chemistry B. It is not designed as a topic test but is instead intended as a document to support the teaching and learning of Module 1 – Development of practical skills in chemistry.

1. A solid organic compound can be purified by recrystallisation.

Which statement(s) about recrystallisation is/are true?

- 1 The organic compound is more soluble in hot solvent.
- 2 The hot solution is cooled before the purified organic compound is collected.
- 3 The melting point of the purified organic compound is lower than the impure compound.

- A** 1, 2 and 3
- B** Only 1 and 2
- C** Only 2 and 3
- D** Only 1

Your answer

[1]

2. When are insoluble impurities removed during recrystallisation?

- A** when the hot solution is filtered
- B** as the solution cools
- C** when the crystals are filtered off
- D** when the crystals are washed

Your answer

[1]

Mark Scheme 2014 Chemistry Ocr F321

Richard Foulston

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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