

NATURAL SCIENCES

Gr. 9 - LESSON PLAN - MATTER & MATERIALS

TOPIC: COMPOUNDS		
Sub-topic: Names of Compounds		Duration: 1 Hour
CONTENT & CONCEPTS (CAPS p63)		
KEY CONCEPTS 1. Chemical reaction 2. Chemical formula 3. Names of compounds according to their elements.		
Specific Aims:	Specific Aim 1: 'Doing Science'	X
	Specific Aim 2: 'Knowing the subject content and making connections'	X
	Specific Aim 3: 'Understanding the uses of Science'	
LESSON OBJECTIVES: 1. Learners will be able to write the names of compounds as indicated on page 63 of the CAPS Document. 2. Learners will be able to make models of different compounds		
RESOURCES REQUIRED: • Periodic Table Poster • Siyavula Sasol Inzalo (Learner book) page 163		
TEACHING & LEARNING ACTIVITY: Introduction <u>Briefly review and revise concepts dealt with in Grade 8, focusing on compounds</u> As each element has a unique name for example: - Hydrogen - Lithium Remember the first 20 elements learnt in grade 8. How did you memorize them? Do you remember Mnemonics... He - H; He - He; Lies - Li; Because - Be; Boys - B; Can - C; Not - N; Operate - O; Fireplaces - F; New - Ne; Nation - Na; Might - Mg; Also - Al; Sign - Si; Peace - P; Security - S; Clause - Cl A - Ar; King - K; Can - Ca LEARNERS MAY RECITE THIS MNEMONIC AS A WAY OF MEMORIZING THE FIRST 20 ELEMENTS. <u>The teacher provides the following explanation to the learners</u> Each compound has a unique name. How is a compound formed? - A compound is formed when TWO or more different elements combine chemically. - Each atom in a compound are held by chemical bonds - Compounds are formed during chemical reaction (when elements/ compounds combine to form new substances) - Each compound can be written as a formula, for example, H₂O; NaCl; CuSO₄ What is a formula ? - This is a notation that shows which elements the compound is made up of. - It indicates the total number of atoms in one unit of the compound - It indicates the ratio in which atoms of different elements are bonded to each other - It gives the total number of atoms in one unit of a compound. For example, the formula for Copper Sulphate is CuSO ₄		

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perspective on development cooperation Part IV provides at a glance comparisons of donors aid outlook and commitment to development cooperation in the 21st Century poverty eradication gender and public support Part V contains handy reference material Throughout the book information is presented in easily interpreted diagrams and graphs The Reality of Aid has established itself as a unique source of independent evaluation and comment on aid policies and development It is indispensable for all in the field whether in the official or voluntary sectors Indispensable it gives you most of the hard facts you need to know about the major issues New Internationalist The most comprehensive and rigorous independent analysis of the aid and development policies of the world s wealthiest nations Essential reading Charity World The Reality of Aid remains an essential purchase by the libraries of development institutions and an invaluable reference for development practitioners Development Change A reliable watchdog for anyone interested in this important aspect of international relations ORBIT An accessible reference it encompasses many key issues and stimulates further research Commonwealth Comparative Politics Originally published in 1998 Membership & NSTA Publications Catalog National Science Teachers Association, 1997

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