

PHYSICAL SCIENCE: Molecular Theory

Learning Objectives

- to identify/understand difference in behaviour of particles/atoms/molecules in different states and understand how it changes as understood in the postulates of the kinetic molecular theory
- to identify/understand/compare different properties of matter
- to represent particle behaviour using simplified kinetic theory of molecular movement

Module 1: Representation of states in Different Phases

The following figure shows the representation of particles in three different states of matter.

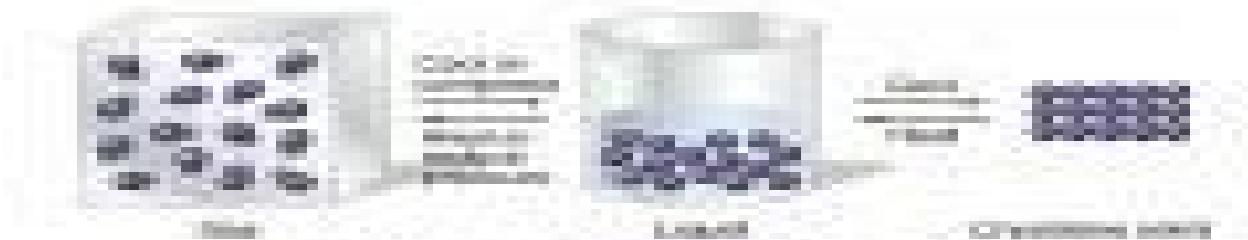


Figure 1: Represented states of matter at different phases of matter

Students might be confused with the representation of particles in different states of matter. Hence, representation with clear and correct meaning.

Key Concepts

- Effect of the type of representation of matter and movement on given matter, and explain the effect of heat on the movement of particles in different states of matter. (as stated in the question, only explain the meaning, the statement of particle movement is not a factor for this question and not the answer)

Table 1: Representation of the three states of matter

	SOLID STATE	LIQUID STATE	GASEOUS STATE
How are the particles arranged?	Particles packed in a regular pattern. Particles are packed closely together. The intermolecular force is high.	Particles packed in a regular pattern. Particles are packed closely together. The intermolecular force is high.	Particles packed in a regular pattern. Particles are packed closely together. The intermolecular force is high.
How movement of particles is in the solid state?	Particles move in a regular pattern. Particles are packed closely together. The intermolecular force is high.	Particles move in a regular pattern. Particles are packed closely together. The intermolecular force is high.	Particles move in a regular pattern. Particles are packed closely together. The intermolecular force is high.

Lw 1 Pogil Kinetic Molecular Theory Key

M Carnoy



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