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Thomas Nelson

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web in this article we will discuss polyatomic ions the prefix poly means many so a polyatomic ion is an ion that contains more than one atom this differentiates polyatomic ions from monatomic ions which contain only one atom examples of monatomic ions include na fe 3 cl and many many others

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web because these ions contain more than one atom they are called polyatomic ions polyatomic ions have characteristic formulas names and charges that should be memorized for example the nitrate ion has one nitrogen atom and three oxygen atoms with an overall charge of 1

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$\textbf{nomenclature for polyatomic ions learnwithdrscott com} \cdot \texttt{Apr} \ 22 \ 2022$

web polyatomic ions are charged groups of atoms an example is ammonium ion nh4 it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using

subscripts for example ammonium chloride is nh4cl and ammonium sulfide is nh4 2s

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web the formulae of compounds containing polyatomic ions are worked out in a similar way to single atom ions except when there is more than one polyatomic ion then its formula is written

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