

| Element           | Q     | R                | V                             | X               | Y                             | Z                |
|-------------------|-------|------------------|-------------------------------|-----------------|-------------------------------|------------------|
| Atomic Radii(nm)  | 0.136 | 0.099            | 0.125                         | 0.117           | 0.110                         | 0.157            |
| Formulae of oxide | QO    | R <sub>2</sub> O | V <sub>2</sub> O <sub>5</sub> | XO <sub>2</sub> | Y <sub>2</sub> O <sub>3</sub> | Z <sub>2</sub> O |
| Melting point(°C) | 650   | 110              | 660                           | 119             | 44.2                          | 97.8             |
| Conductivity      | Good  | Poor             | Good                          | Poor            | Poor                          | Good             |

- (i) From the table which elements could be:
- I. Magnesium.....(1 mark)
- II. Sulphur.....(1 mark)
- (ii) Write the formula of the chloride of X. ....(1 mark)
- (iii) Arrange the elements as they appear from left to right in the periodic table  
(2marks)
- .....
- .....
- (iv) What type of bonding would you expect the chloride of Y to exhibit? (1 mark)
- .....
- .....
- (v) Write a chemical equation for the reaction of oxide of Y with water. (1 mark)
- .....
- .....

(b) Use the information in the table below to answer the questions that follow.

| Element | Ionization Energy in kJ |                 |                 |                 |
|---------|-------------------------|-----------------|-----------------|-----------------|
|         | 1 <sup>st</sup>         | 2 <sup>nd</sup> | 3 <sup>rd</sup> | 4 <sup>th</sup> |
| V       | 320                     | 580             | 4900            | 7200            |
| W       | 430                     | 4300            | 6200            | 8300            |
| X       | 600                     | 1050            | 1450            | 16000           |
| Y       | 7000                    | 9000            | 10500           | 14000           |
| Z       | 200                     | 420             | 5200            | 6600            |

- (i) Name two elements that can be found in the same group? (1 mark)
- .....
- .....
- (ii) To which group does element W belong? Explain. (2marks)
- .....
- .....

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