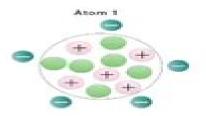
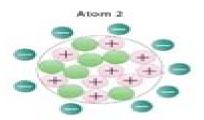
- 1. Write the symbol for the following isotopes:
- a. 8 protons and 8 neutrons

c. atomic number is 11 and mass number is 23

b. 28 protons and 30 neutrons

- d. 92 protons and mass number is 238
- 2. Fill in the table for the two atoms shown below.





	Atom 1	Atom 2
Number of Protons		
Number of Neutrons		
Mass Number		

3. Fill in the following table.

Isotope Symbol	40 19	18 9F	
Atomic Number			16
Mass Number	· · · · · · · · · · · · · · · · · · ·		
Number of Protons	>		
Number of Neutrons			15

Name:

Date : _____

Isotopes Practice Set

Edith Hinkley Quimby

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