

Mole Ratios POGIL Part 2

What mass of nitrogen is needed to produce 30.0 g of ammonia?

we have : 30.0 g of ammonia (NH_3)

we want : grams of nitrogen (N_2)



Step 1: get moles! (use molar mass)

$$\frac{30.0 \text{ g NH}_3}{17.04 \text{ g NH}_3} \times 1 \text{ mol NH}_3 = 1.76 \text{ mol NH}_3$$

More Ratios Pogil Answer

Rachel Sandford



More Ratios Pogil Answer:

Reviewing **More Ratios Pogil Answer**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**More Ratios Pogil Answer**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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