

1. For the formation of double helical DNA from its single strands in a solution, which of the following is not applicable ?
 - (A) The entropy of the surrounding is increased
 - (B) The entropy of the solution is increased
 - (C) Appreciable enthalpy change results
 - (D) Does not obey second law of thermodynamics
2. Biological systems make the thermodynamically unfeasible reactions possible by :
 - (A) Manipulating the concentrations of products and reactants such that free energy change favors the reaction to occur
 - (B) Coupling them to the hydrolysis of ATP
 - (C) Coupling them with exergonic reactions via a common intermediate
 - (D) All of the above
3. The entropy may be expressed as a function of :
 - (A) Pressure and temperature
 - (B) Temperature and volume
 - (C) Heat and work
 - (D) All of the above
4. On passing electric current through aqueous solutions of the following substances, which one shall decompose ?
 - (A) Urea
 - (B) AgNO_3
 - (C) Glucose
 - (D) Ethyl alcohol
5. pH of a solution is defined as $\text{pH} = -\log [\text{H}^+]$, where $[\text{H}^+]$ is in units of :
 - (A) Molarity
 - (B) Molality
 - (C) Normality
 - (D) Normality or Molality
6. Single strands of a DNA molecule separate at :
 - (A) High pH
 - (B) Low pH
 - (C) High temperature
 - (D) All of the above
7. Most of the important functional groups in biological molecules contain :
 - (A) Oxygen and a phosphate
 - (B) Oxygen and/or nitrogen and are acidic
 - (C) Nitrogen and a phosphate
 - (D) Oxygen and/or nitrogen and are polar

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