

6. Polystyrene	Styrene	$\text{CH}_2=\text{CH}$ C_6H_5
7. Buna rubber	Buta-1,3-diene	$\text{CH}_2=\text{CH}-\text{CH}=\text{CH}_2$
8. Dacron (Terylene)	Ethane-1,2-diol + 1,4-benzenedioic acid	CH_2-CH_2 OH OH + $\text{HO}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{C}_6\text{H}_4)-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$
9. Nylon 6,6	1,6-diamino hexane + 1,6-hexanedioic acid	$\text{NH}_2-(\text{CH}_2)_6-\text{NH}_2$ + $\text{HO}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_4-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$ $\text{C}_6\text{H}_5-\text{OH}$ + HCHO
10. Bakelite	Phenol + formaldehyde	

2. On the basis of monomers.

a. Homopolymers

→ It is formed by only one type of monomer unit.

For eg:- polythene, polyvinyl chloride (PVC), Orlon, Teflon etc.

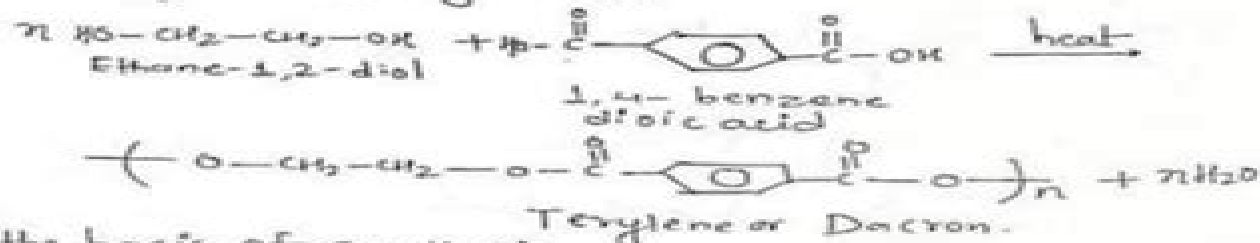


b. Copolymers:

→ It is formed by more than one type of monomers

→ Monomers are arranged in definite structural pattern in a random manner.

For eg Dacron, Nylon 6,6



3 On the basis of synthesis.

a. Addition polymers

→ Large number of monomer molecules are added together to form high molecular weight

For eg:- pvc, orlon, te flon

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Klencsár, Rezso György Lovas, Frank
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Table of Contents Mcq On Nuclear Chemistry

1. Understanding the eBook Mcq On Nuclear Chemistry
 - The Rise of Digital Reading Mcq On Nuclear Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcq On Nuclear Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcq On Nuclear Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcq On Nuclear Chemistry
 - Personalized Recommendations

- Mcq On Nuclear Chemistry User Reviews and Ratings
- Mcq On Nuclear Chemistry and Bestseller Lists
- 5. Accessing Mcq On Nuclear Chemistry Free and Paid eBooks
 - Mcq On Nuclear Chemistry Public Domain eBooks
 - Mcq On Nuclear Chemistry eBook Subscription Services
 - Mcq On Nuclear Chemistry Budget-Friendly Options
- 6. Navigating Mcq On Nuclear Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Mcq On Nuclear Chemistry Compatibility with Devices
 - Mcq On Nuclear Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcq On Nuclear Chemistry
 - Highlighting and Note-Taking Mcq On Nuclear Chemistry
 - Interactive Elements Mcq On Nuclear Chemistry
- 8. Staying Engaged with Mcq On Nuclear Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcq On Nuclear Chemistry
- 9. Balancing eBooks and Physical Books Mcq On Nuclear Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcq On Nuclear Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mcq On Nuclear Chemistry
 - Setting Reading Goals Mcq On Nuclear Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mcq On Nuclear Chemistry
 - Fact-Checking eBook Content of Mcq On Nuclear Chemistry

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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