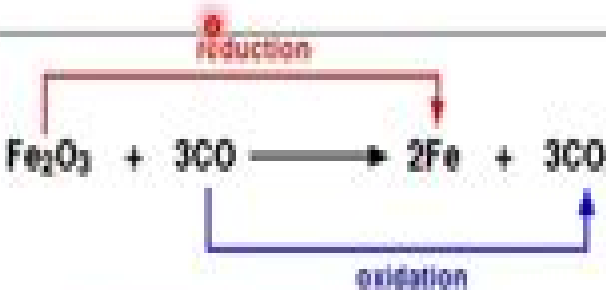
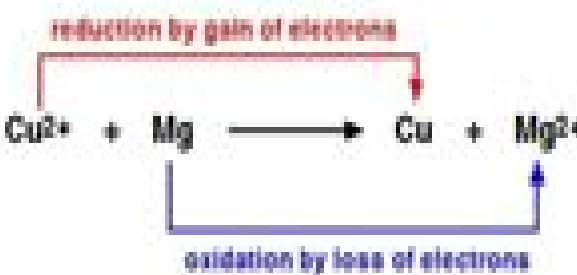
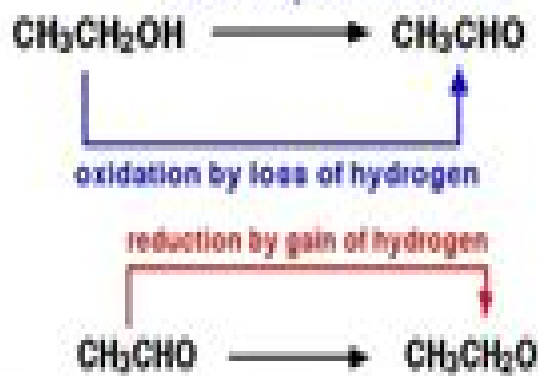


Definitions of Oxidation and Reduction

Definition No.	Oxidation	Reduction	Example
1	gain of oxygen	loss of oxygen	 $\text{Fe}_2\text{O}_3 + 3\text{CO} \longrightarrow 2\text{Fe} + 3\text{CO}_2$
2	loss of electrons	gain of electrons	 $\text{Cu}^{2+} + \text{Mg} \longrightarrow \text{Cu} + \text{Mg}^{2+}$
3	loss of hydrogen	gain of hydrogen	 $\text{CH}_3\text{CH}_2\text{OH} \longrightarrow \text{CH}_3\text{CHO}$ $\text{CH}_3\text{CHO} \longrightarrow \text{CH}_3\text{CH}_2\text{OH}$



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atomic configuration of real solids and on the relationship of composition and structure to the chemical and physical properties of the solid Real solids are usually extremely complex and exhibit almost infinite variety in their compositional and structural features

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