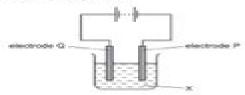
Chemistry Past Papers

2015 M P22

8 The diagram snows an electrolysis experiment.



During the electrolysis, sodium was formed at electrode P and officine at electrode Q.

Which row correctly identifies P. Q and X7.

	p-	-0	N
	amode	cathode	concentrated solution of sodium chloride in water
	anode	cathode	molten sodium chloride
C	cathode	enode	concentrated solution of sodium chloride in water
0	similarida:	amode:	molten sedium chloride

12 Hydrated cobatt(II) chloride decomposes on heating.

The equation for the reaction is

The reaction is reversed by adding water.

Which row describes the opiour change and the type of reaction for the reverse reaction?

	colour change	type of reaction
A	blue to pink	endothermic
0	blue to pink	exothermic
C	pink to blue	endothermic
D.	prink to blue	exothermic

- 4 (a) A compound X contains 82.76% of carbon by mass and 17.24% of hydrogen by mass.
 - (i) Calculate the empirical formula of compound X.

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M Walker

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