

## Chemistry -- Unit 6 Worksheet 1

After having observed molecular models which show 18.098 compounds, you discovered that certain particles are known to form positively charged ions, but cannot, within their natural particles, form negatively charged ions. (Lithium, Sodium, Potassium, Rubidium, Cesium, Francium, Barium, Strontium, Calcium, Magnesium, Beryllium, and Radium). After having observed the 18.098 compounds, you discovered that certain particles are known to form negatively charged ions, but cannot, within their natural particles, form positively charged ions. (Fluorine, Chlorine, Bromine, Iodine, Astatine, and Tellurium). After having observed the 18.098 compounds, you discovered that certain particles are known to form positively charged ions, but cannot, within their natural particles, form negatively charged ions. (Lithium, Sodium, Potassium, Rubidium, Cesium, Francium, Barium, Strontium, Calcium, Magnesium, Beryllium, and Radium).

1. After having observed molecular models which show 18.098 compounds, you discovered that certain particles are known to form positively charged ions, but cannot, within their natural particles, form negatively charged ions. (Lithium, Sodium, Potassium, Rubidium, Cesium, Francium, Barium, Strontium, Calcium, Magnesium, Beryllium, and Radium).

18.098 molecular formulas	1. Lithium 1. Sodium	1. Potassium 1. Rubidium	1. Calcium 1. Barium	1. Strontium 1. Radium
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positive  
charge



18.098 molecular formulas	1. Lithium 1. Sodium	1. Potassium 1. Rubidium	1. Calcium 1. Barium
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18.098 molecular formulas	1. Lithium 1. Sodium	1. Potassium 1. Rubidium	1. Calcium 1. Barium	1. Strontium 1. Radium
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positive  
charge



# Modeling Chemistry U6 Ws1 V2 0 Answers

**Xiaolong Qi**



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